

# Variational solution of the Schrödinger equation in an inhomogeneous central field as applied to emission problems.

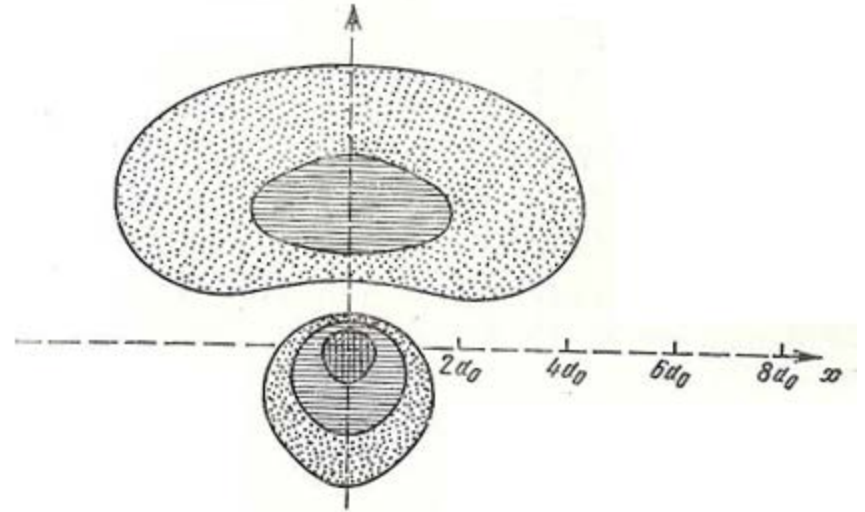
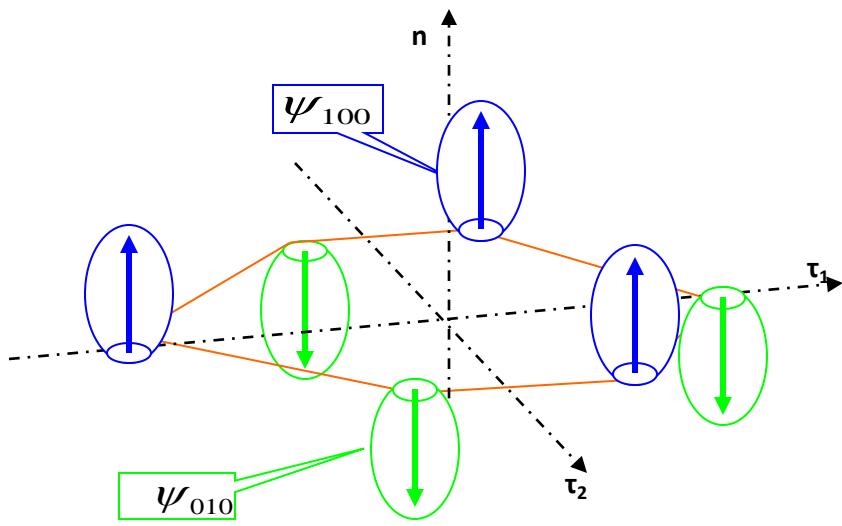
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Functional for determining the effective charge for the model ion potential

$$J_1(q) = \langle R_s^*(x) U_i(r) R_s \rangle \quad x = \frac{2q}{n} r$$

Hydrogen-like atom model :

$$U(r) = \frac{q_{eff}}{r}$$

The operator of potential energy in a homogeneous field [W. Brandt, M. Kitagawa, Phys. Rev. B,1982,v.25,p. 5631-5637]

$$U(\mathbf{r}) = \frac{Z}{r} - \frac{Z-1}{r} \left[ 1 - \exp\left(-\frac{r}{\lambda}\right) \right] = \frac{1}{r} + \frac{Z-1}{r} \exp\left(-\frac{r}{\lambda}\right)$$

The functional for determining the effective charge  $q$  for a homogeneous field

$$J(q) = -\frac{q^2}{2n^2} + \int_0^{\infty} R_{ns}^2(r, q) V(r) r^2 dr + \frac{q^2}{n^2}$$

# Variational method

[Nikiforov, Novikov, Uvarov]

$$\varepsilon_0 = \min \int \psi^* \hat{H} \psi d\xi$$

$$J(\alpha, \beta, \dots) = \int \psi^*(\xi, \alpha, \beta, \dots) \hat{H} \psi(\xi, \alpha, \beta, \dots) d\xi.$$

$$\frac{\partial J}{\partial \alpha} = 0, \quad \frac{\partial J}{\partial \beta} = 0, \quad \dots ,$$

The Brandt-Kitagawa potential is inhomogeneous in radius

$$\frac{\partial}{\partial r} U_i(r) = -\frac{U_i(r)}{r} - \frac{Z-1}{\lambda_i} \exp\left(-\frac{r}{\lambda_i}\right) \neq \chi \frac{U_i(r)}{r}$$

For the inhomogeneous field of the ion, it is necessary to take into account the full energy of the electron by using Virial theorem

$$J_2(q) = \langle R_s^*(x) \left[ \frac{r}{2} \frac{\partial}{\partial r} U_i(r) + U_i(r) \right] R_s \rangle \quad x = \frac{2q}{n} r$$

Objective: finding an effective uniform Coulomb field of the form

$$U(r) = -\frac{q}{r} + A \quad \text{with ground-state energy} \quad E_s = \frac{q^2}{2n^2}$$

Let's minimize error between full energy, found with Brandt-Kitagawa potential and homogeneous one

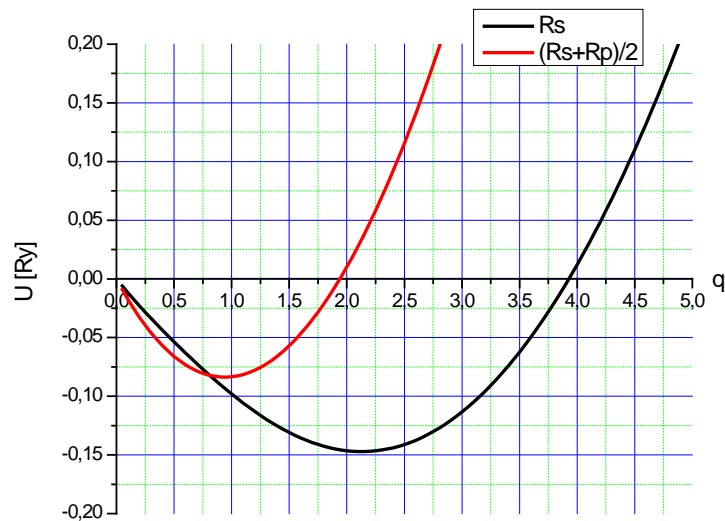
$$\frac{\partial}{\partial q} \left\{ \langle R_s(r) \left[ \frac{r}{2} \frac{\partial}{\partial r} U_i(r) + U_i(r) \right] R_s(r) \rangle - \frac{q^2}{2n^2} \right\} = 0$$

After solution of variation problem we can find external screening potential

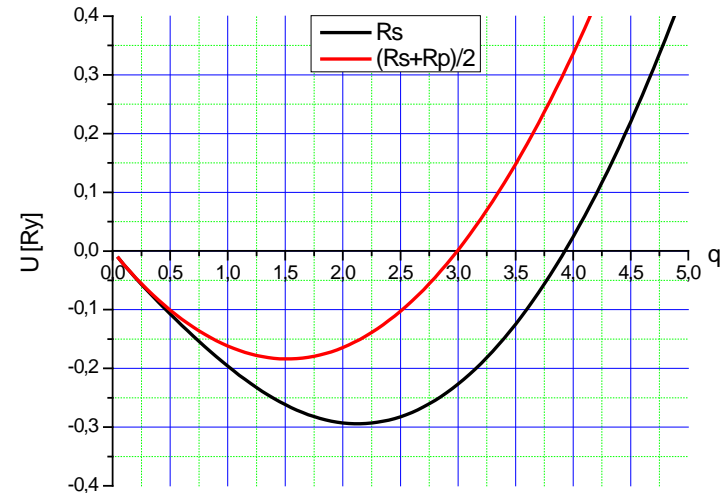
$$A = \langle R_s(r) \left[ U_i(r) - \frac{q}{r} \right] R_s(r) \rangle$$

# Comparison of solutions of the variational problem

Average potential energy



Average full energy



$R_s$

$$(R_s + R_p \cos \theta) / 2$$

$$q \quad 2.130985 \quad 0.9417036$$

$$A \quad -0.1471367 \quad -8.3710894E-$$

02

$R_s$

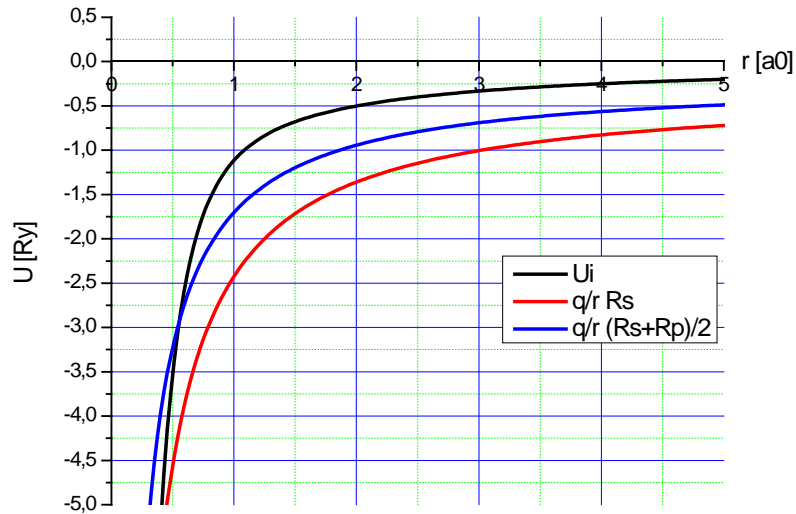
$$(R_s + R_p \cos \theta) / 2$$

$$q \quad 2.130985 \quad 1.519652$$

$$A \quad -0.2942734 \quad -0.1839328$$

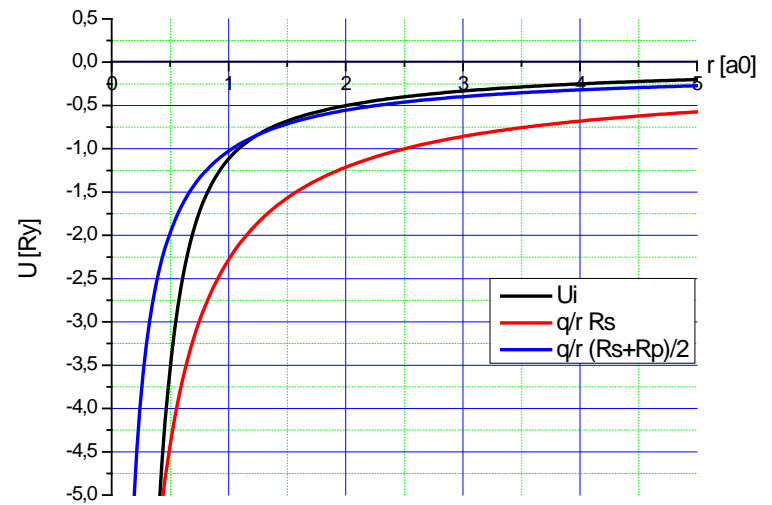
# Comparison of the initial and effective fields

Average potential energy



$R_s$   
 $(R_s + R_p \cos \theta) / 2$   
 $q$  2.130985 0.9417036  
 $A$  -0.1471367 -8.3710894E-02

Average full energy



$R_s$   
 $(R_s + R_p \cos \theta) / 2$   
 $q$  2.130985 1.519652  
 $A$  -0.2942734 -0.1839328



## Conclusion:

Accounting of heterogeneity allows you to specify a solution variational problem in a model of a hydrogen atom

These studies will help to build a more accurate adaptive model of the atom in the lattice, which takes into account the field of environment and external sources

The model of the hydrogen-like atom can be used to solve various problems using multiprocessor systems

Thank you for attention!