



### PHOTOPROTON REACTIONS ON NATURAL MIXTURE OF STRONTIUM

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#### *s*-process – slow neutron capture

*r*-process – fast neutron capture

*p*-process – sequential photonuclear reactions ( $\gamma$ ,1*n*) or multiparticle reactions ( $\gamma$ ,2*n*)



*Figure.1.* The trajectory of the *s*-process is a wide white line, the *r*-process is a thin white line for isotopes Kr, Rb and Sr Z=36-38



*Figure 2.* Scheme of the experiment

 $^{84}$ Sr (0.56%) +  $^{86}$ Sr (9.86%) +  $^{87}$ Sr (7.00%) +  $^{88}$ Sr (82.58%)



*Figure 3a*. Spectra of the residual activity of the irradiated strontium target for 4 min and 1.58 h after the end of irradiation. The duration of spectrum measurements was 30 min (short irradiation)



*Figure 3b.* Spectra of the residual activity of the irradiated strontium target for 2.83 hours and 3.04 days after the end of irradiation. The duration of the spectra measurements was 3 hours

$$y_{1} = \frac{N_{10}}{e^{-\lambda_{1}t_{1}} \sum_{i=0}^{t} I(t_{i}) \cdot e^{\lambda_{1}t_{i}} \Delta t} \quad (1) \qquad N_{10} = \frac{S}{k \cdot (e^{-\lambda_{1}(t_{2}-t_{1})} - e^{-\lambda_{1}(t_{3}-t_{1})})} \quad (2)$$

where

 $N_{10}$  — number of nuclei at the end of irradiation

 $\lambda$  — decay constant

S — photopeak area

I(t) — accelerator current

 $t_1$  — irradiation time

 $t_2$  — start time of spectra measurement

 $t_3$  — end time of spectrum measurement

k — coefficient taking into account cascade transitions and the effect of self-absorption in the target ( $k=\varepsilon \cdot I_{\gamma} \cdot T_{cc}$ )

# *Table 1.* Experimental yields of ${}^{nat}Sr(\gamma,xn1p)$ reaction products produced using bremsstrahlung endpoint energy of 55 MeV

| N⁰ | A                 | E, keV ( $I_{\gamma}$ , %)                      | $Y_{exp}, 1/e$                       | T <sub>1/2</sub> |
|----|-------------------|---|--------------------------------------|------------------|
| 1  | <sup>81</sup> Rb  | 190.46 (64.9), 446.15 (23.5)                    | $(1.44 \pm 0.11) \cdot 10^{-9}$      | 4.57 h           |
| 2  | <sup>82m</sup> Rb | 554.35 (62.4), 619.11 (37.98), 698.37 (26.3),   | $(5.01 \pm 0.59)$ ·10 <sup>-10</sup> | 6.47 h           |
|    |                   | 776.52 (84.39), 827.83 (21), 1007.59 (7.17),    |                                      |                  |
|    |                   | 1044.08 (32.07), 1317.43 (23.7), 1474.88 (15.5) |                                      |                  |
| 3  | <sup>83</sup> Rb  | 520.39 (45), 529.59 (29.3), 552.55 (16)         | $(3.34 \pm 0.26)$ ·10 <sup>-8</sup>  | 86.2 d           |
| 4  | <sup>84</sup> Rb  | 881.60 (68.9)                                   | $(1.22 \pm 0.39)$ ·10 <sup>-8</sup>  | 32.77 d          |
| 5  | <sup>86</sup> Rb  | 1077.0 (8.64)                                   | $(1.06 \pm 0.04)$ ·10 <sup>-8</sup>  | 18.63 d          |
| 6  | <sup>86m</sup> Rb | 556.07 (98.21)                                  | $(2.48 \pm 0.04)$ ·10 <sup>-8</sup>  | 1.017 min        |

The theoretical yield of isotope formation, taking into account all possible reactions leading to the formation of the selected isotope, was calculated by Equation (3):

$$Y = \lambda \alpha \sum_{i} \eta_{i} \int_{E_{i}}^{E_{m}} \phi(E_{\gamma}, E_{m}) \sigma_{i}(E_{\gamma}) dE_{\gamma}$$
(3)

where

 $\lambda$  — a decay constant,  $\alpha$  is the number of studied nuclei per 1 cm<sup>2</sup> of target;

index *i* corresponds to the number of the reaction contributing to the formation of studied isotope;

 $\eta_i$  — the percentage of the strontium isotope on which the reaction occurs in a natural mixture of isotopes;

 $E_i$  — the threshold of the corresponding reaction,  $E_m$  is the maximum energy of the bremsstrahlung spectrum;

 $\sigma_i(E_{\gamma})$  — the cross-section of the corresponding photoproton reaction;

 $\varphi(E_{\gamma}, E_m)$  — the bremsstrahlung spectrum on the target.

*Table 2.* Comparison of experimental  $Y_{exp}$  and theoretical yields of photonuclear reactions on strontium isotopes, calculated on the basis of TALYS ( $Y_{TALYS}$ ). For partial reactions, the second column gives a coefficient equal to the content of stable selenium isotopes in the natural mixture, on which reactions occur, leading to the formation of the studied isotopes

| N⁰ | А                 | Reaction                                  | E <sub>th</sub> , MeV | Yields, 1/e            |                                      |
|----|-------------------|---|-----------------------|------------------------|--------------------------------------|
|    |                   |   |                       | TALYS-1.6              | Experiment                           |
| 1  | <sup>81</sup> Rb  | $^{nat}Sr(\gamma,in1p) =$                 |                       | 1.12.10-9              | $(1.44 \pm 0.11)$ ·10 <sup>-9</sup>  |
|    |                   | $0.0056^{.84}$ Sr( $\gamma$ ,2n1p) +      | 28.71                 | 5.85·10 <sup>-10</sup> |                                      |
|    |                   | $0.0986^{.86}$ Sr( $\gamma$ ,4n1p)        | 48.73                 | 5.39·10 <sup>-10</sup> |                                      |
| 2  | <sup>82m</sup> Rb | $^{nat}Sr(\gamma,in1p) =$                 |                       | 6.34·10 <sup>-10</sup> | $(5.01 \pm 0.59)$ ·10 <sup>-10</sup> |
|    |                   | $0.0056^{\cdot 84}$ Sr( $\gamma$ ,1n1p) + | 19.81                 | 3.95·10 <sup>-10</sup> |                                      |
|    |                   | 0.0986 <sup>.86</sup> Sr(γ,3n1p)          | 39.91                 | $2.37 \cdot 10^{-10}$  |                                      |
| 3  | <sup>83</sup> Rb  | $^{nat}Sr(\gamma,in1p) =$                 |                       | 0.99.10-8              | $(3.34 \pm 0.26)$ ·10 <sup>-8</sup>  |
|    |                   | $0.0056^{.84}$ Sr( $\gamma$ ,1p) +        | 8.86                  | 9.63·10 <sup>-9</sup>  |                                      |
|    |                   | $0.0986^{.86}$ Sr( $\gamma$ ,2n1p) +      | 28.88                 | $2.58 \cdot 10^{-10}$  |                                      |
|    |                   | $0.07^{.87}$ Sr( $\gamma$ ,3n1p)          | 36.92                 | $2.24 \cdot 10^{-10}$  |                                      |
| 4  | <sup>84</sup> Rb  | $^{nat}Sr(\gamma,in1p) =$                 |                       | 1.45.10-8              | $(1.22 \pm 0.39)$ ·10 <sup>-8</sup>  |
|    |                   | $0.0986^{.86}$ Sr( $\gamma$ ,1n1p) +      | 20.13                 | 9.48·10 <sup>-9</sup>  |                                      |
|    |                   | $0.07 \cdot {}^{87}Sr(\gamma, 2n1p) +$    | 28.64                 | $1.24 \cdot 10^{-9}$   |                                      |
|    |                   | $0.8258^{.88}$ Sr( $\gamma$ ,3n1p) +      | 40.14                 | 3.79·10 <sup>-9</sup>  |                                      |
| 5  | <sup>86</sup> Rb  | $^{nat}Sr(\gamma,in1p) =$                 |                       | 1.07.10-7              | $(3.56 \pm 0.04)$ ·10 <sup>-8</sup>  |
|    |                   | $0.07 \cdot {}^{87}Sr(\gamma, 1p) +$      | 18.07                 | 1.00.10-8              |                                      |
|    |                   | $0.8258^{.88}$ Sr( $\gamma$ ,1n1p)        | 29.19                 | 9.67·10 <sup>-8</sup>  |                                      |

## Conclusions

The method of induced activity was used to study photonuclear reactions with a natural mixture of strontium isotopes. The experiment was carried out on the bremsstrahlung of the RM-55 electron accelerator at an electron energy of 55 MeV. The study studied the possibility of obtaining the <sup>82</sup>Sr isotope in photonuclear reactions on a natural mixture of strontium isotopes.

There are no experimental data on the cross sections for photoproton reactions on Sr isotopes in the literature. The formation yields of rubidium isotopes  $^{81,82m,83,84,86,86m}$ Rb as a result of  $^{nat}$ Sr( $\gamma$ ,*in1p*) reactions were measured. The experimentally obtained yields of photonuclear reactions are compared with the yields calculated using the theoretical cross sections for photonuclear reactions in TALYS-1.6. The difference in the experimental yields and the theoretical calculations values can be due to two main factors: TALYS uses default photoabsorption cross-sections, and also does not take into account the isospin splitting of the giant dipole resonance, which has a significant effect on the yields of photoproton reactions.

# THANK YOU FOR ATTENTION!







*Figure 4*. Change in the average current of the electron beam of the accelerator: a) long irradiation; b) short irradiation