



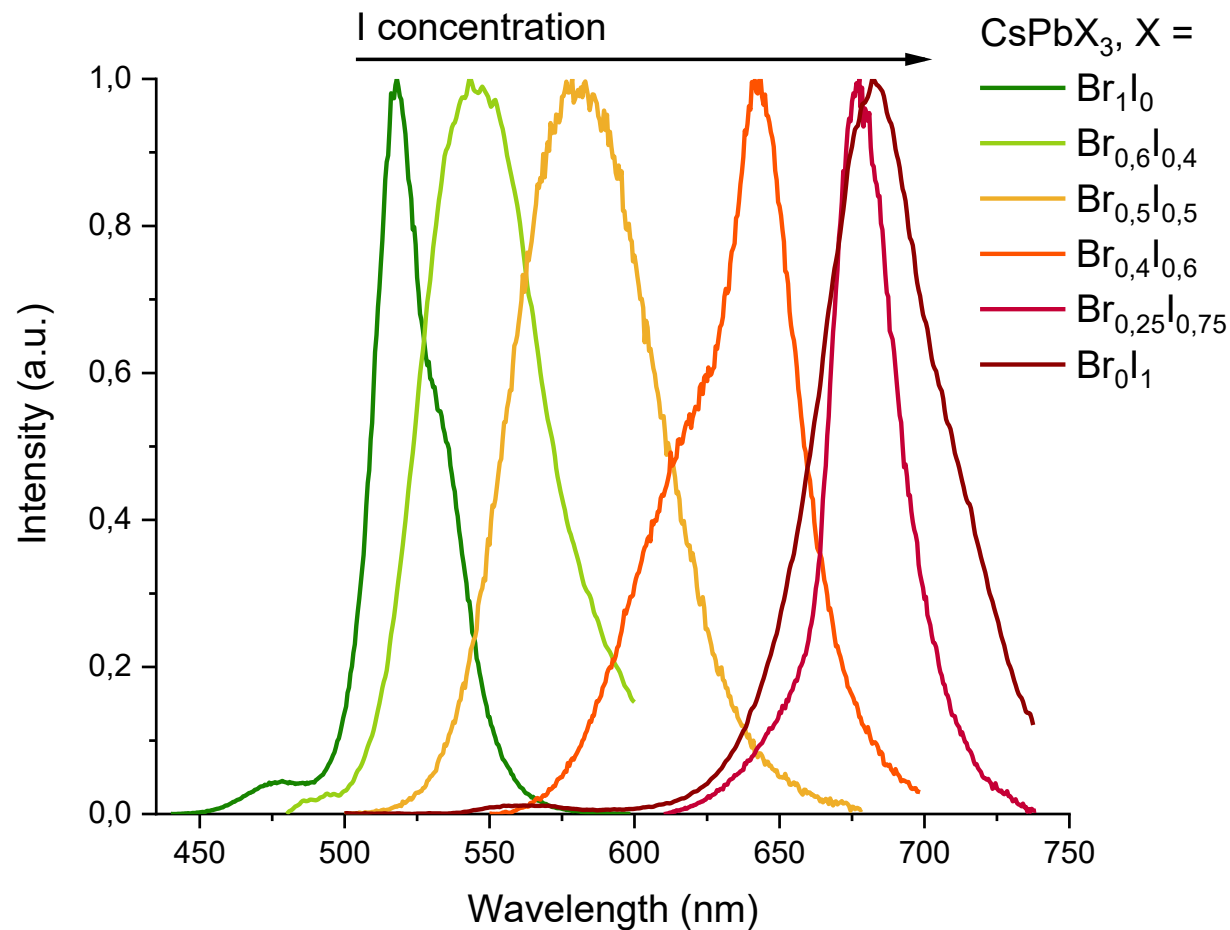
**ITMO**

Optical study of phase transitions of mixed  
perovskite nanocrystals  $\text{CsPb}(\text{Br}_x\text{I}_{1-x})_3$  obtained in  
borogermanate glass

Rufina D. Kharisova, A.N. Babkina, K.S. Zyryanova

Research Center for Optical Material Engineering  
ITMO University, St. Petersburg

# Nanocrystals $\text{CsPbHal}_3$

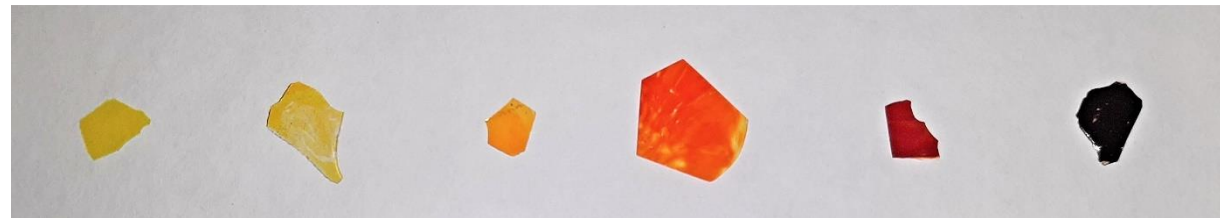


Luminescence spectra

- direct band gap semiconductors,
  - high absorption and emission cross sections,
  - low lasing thresholds,
  - narrow luminescence bands,
  - possibility to tune the luminescence throughout the entire visible spectrum
- 
- perovskites undergo several phase transitions in the temperature range of 30 - 330 °C

# Object of study

- Glass matrix  
 $23,59 \text{ B}_2\text{O}_3 - 38,09 \text{ GeO}_2 - 6,41 \text{ Na}_2\text{O} - 5,03 \text{ ZnO} - 1,38 \text{ P}_2\text{O}_5 - 2,85 \text{ TiO}_2 - 4,99 \text{ K}_2\text{O} - 5,41 \text{ Cs}_2\text{O} - 2,26 \text{ PbO} - 9,98x \text{ Br} - 9,98(1-x) \text{ I mol.}\%$
- $x = 1; 0,6; 0,5; 0,4; 0,25; 1$
- Nanocrystals  $\text{CsPb}(\text{Br}_x\text{I}_{1-x})_3$   
 nucleated during the annealing
- Glass synthesis  
 30 min,  $950 \text{ }^\circ\text{C}$
- $T_g = 440 - 450 \text{ }^\circ\text{C}$



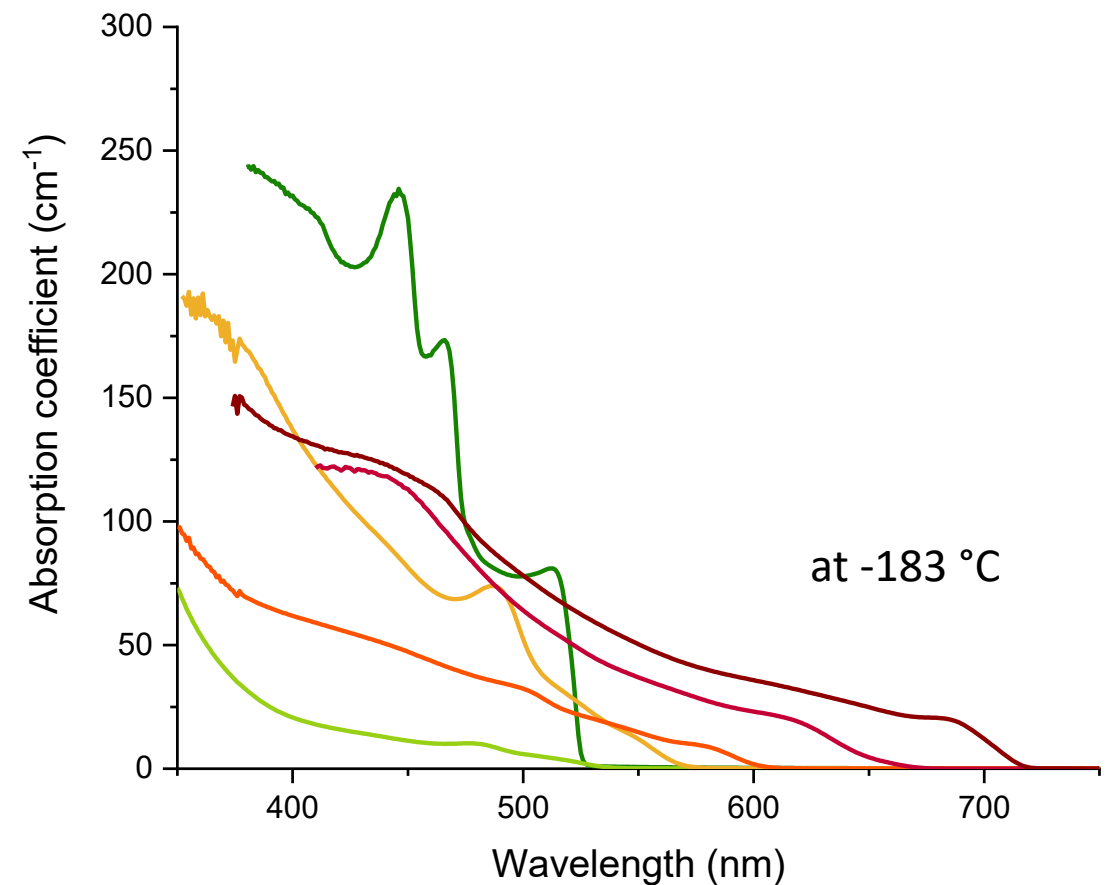
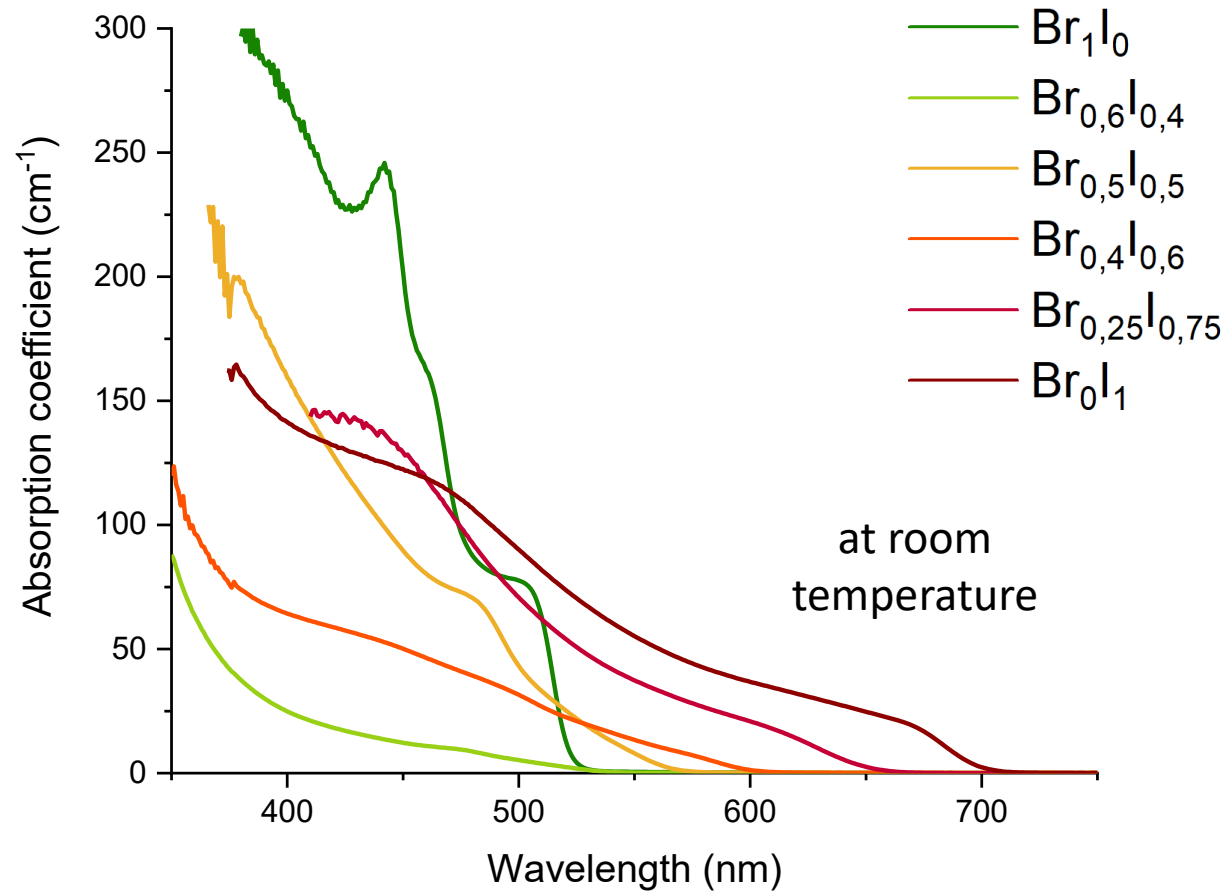
under daylight



under UV irradiation

$\text{Br}_1\text{I}_0$     $\text{Br}_{0,6}\text{I}_{0,4}$     $\text{Br}_{0,5}\text{I}_{0,5}$     $\text{Br}_{0,4}\text{I}_{0,6}$     $\text{Br}_{0,25}\text{I}_{0,75}$     $\text{Br}_0\text{I}_1$

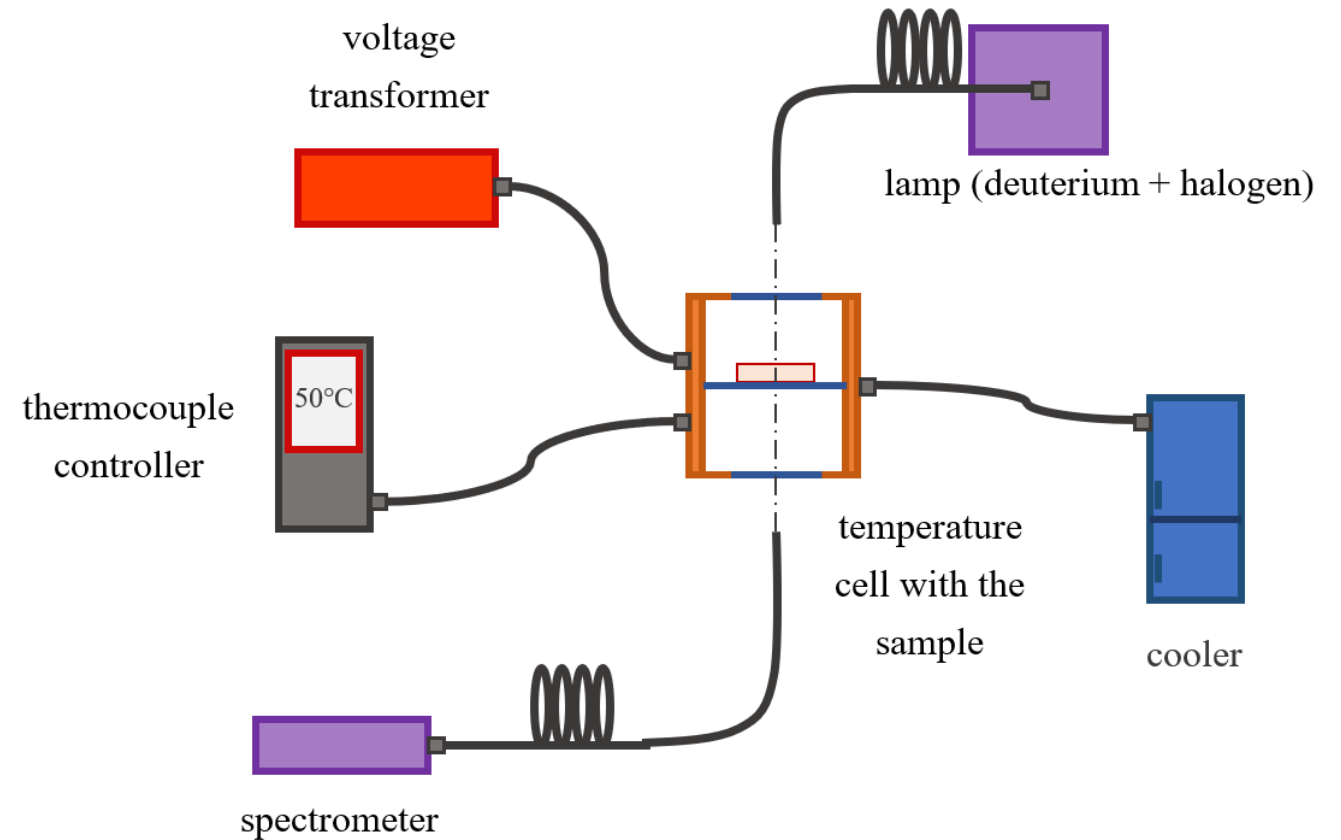
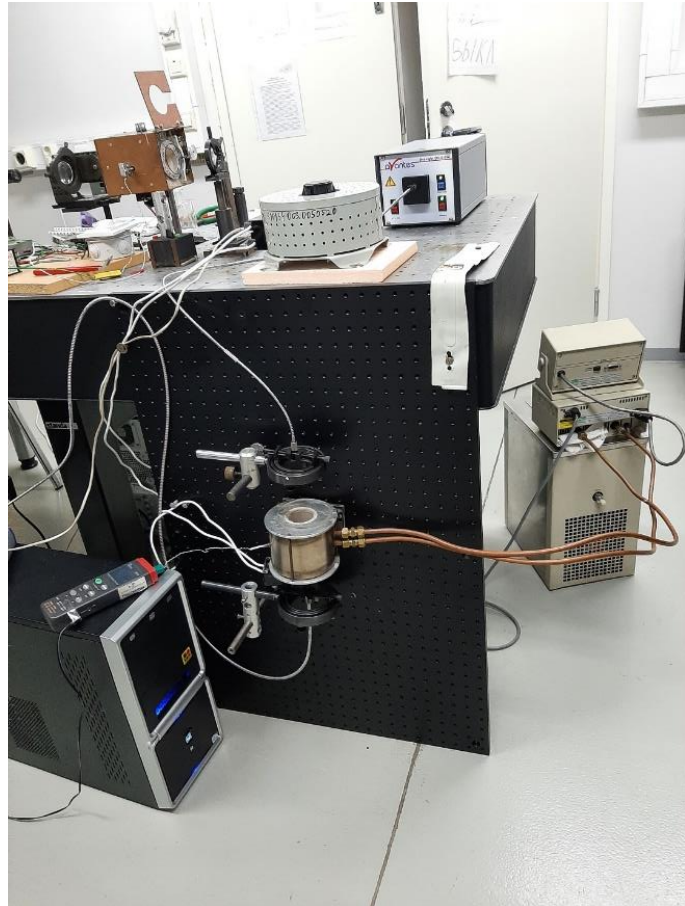
# Absorption spectra



$$\hbar\omega = E_g + 0,71 \frac{\hbar^2 \pi^2}{2\mu a^2}$$

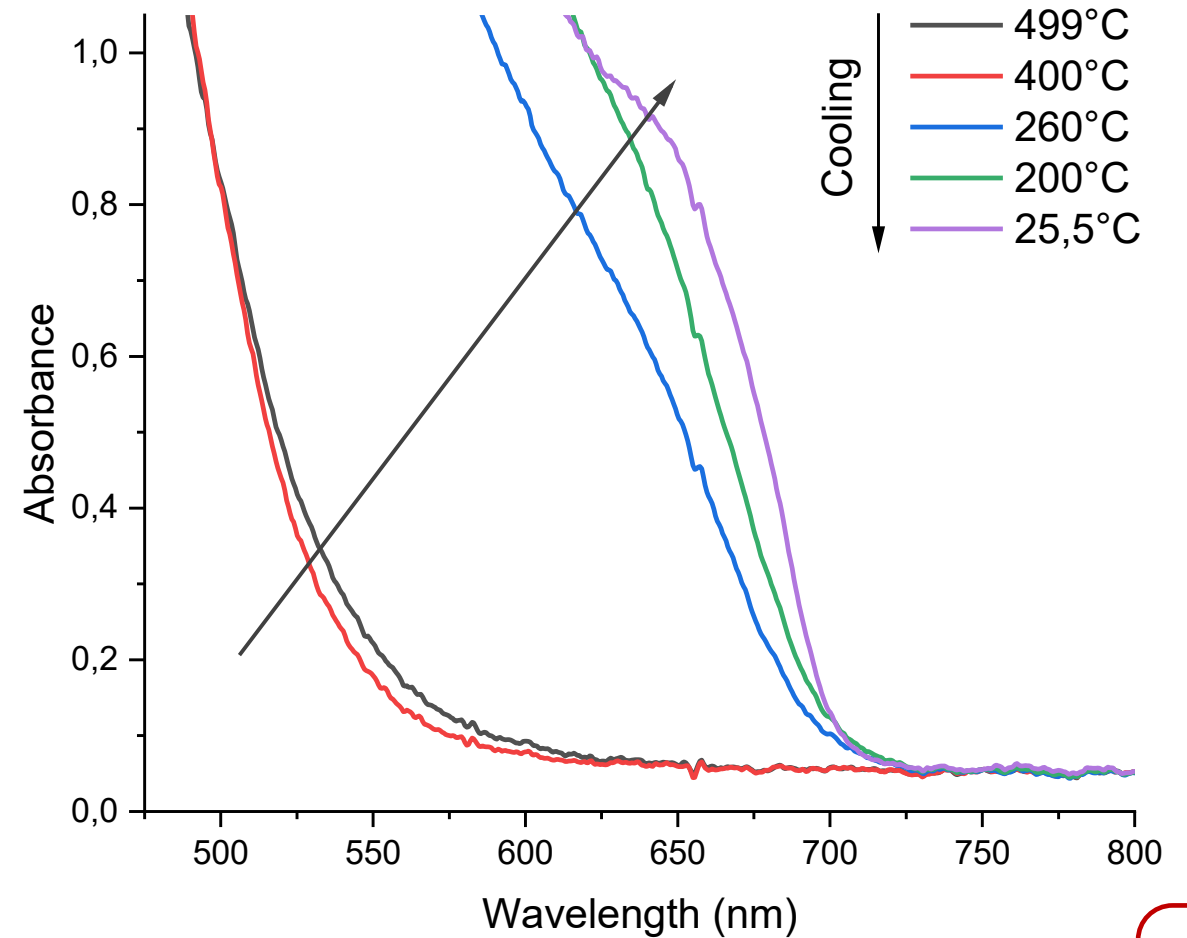
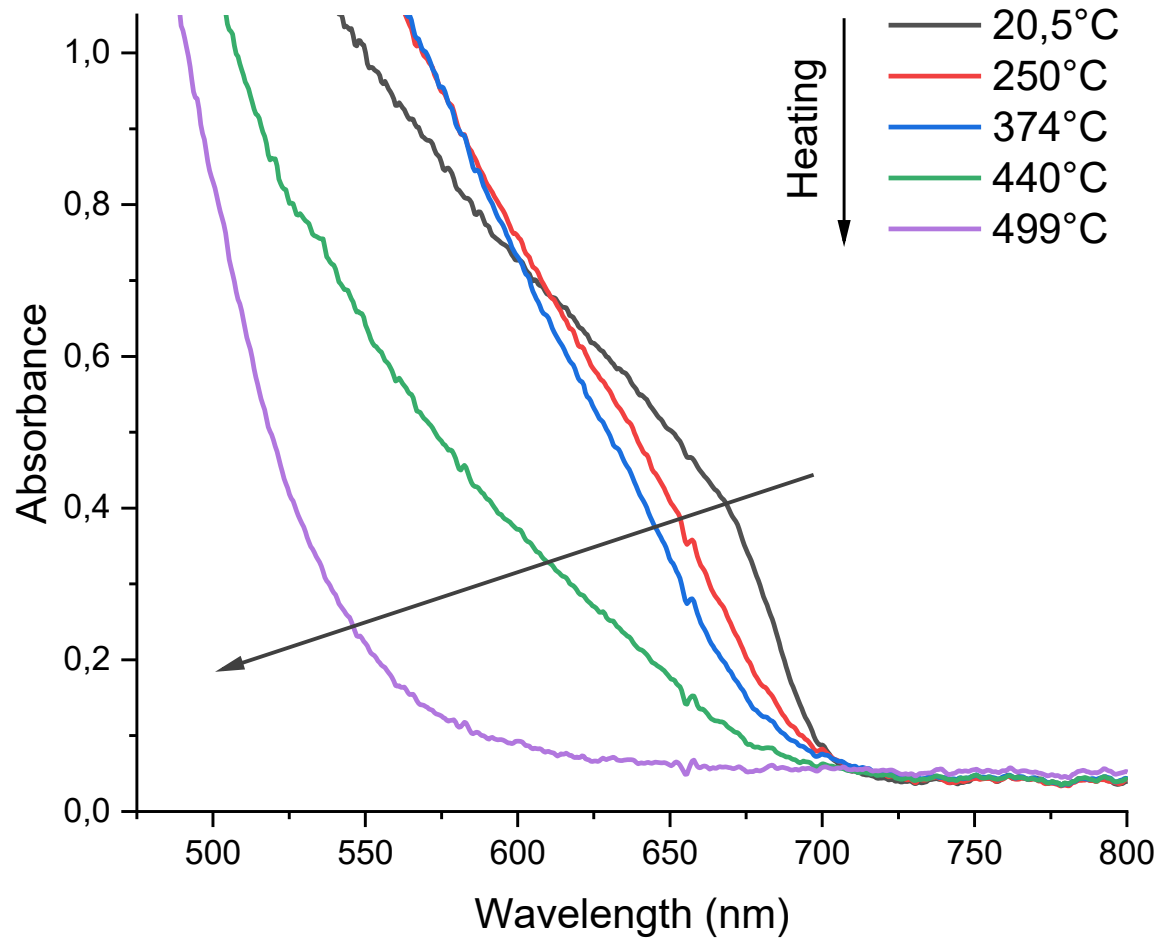
Nanocrystals size 10 – 15 nm

# Temperature measurement setup



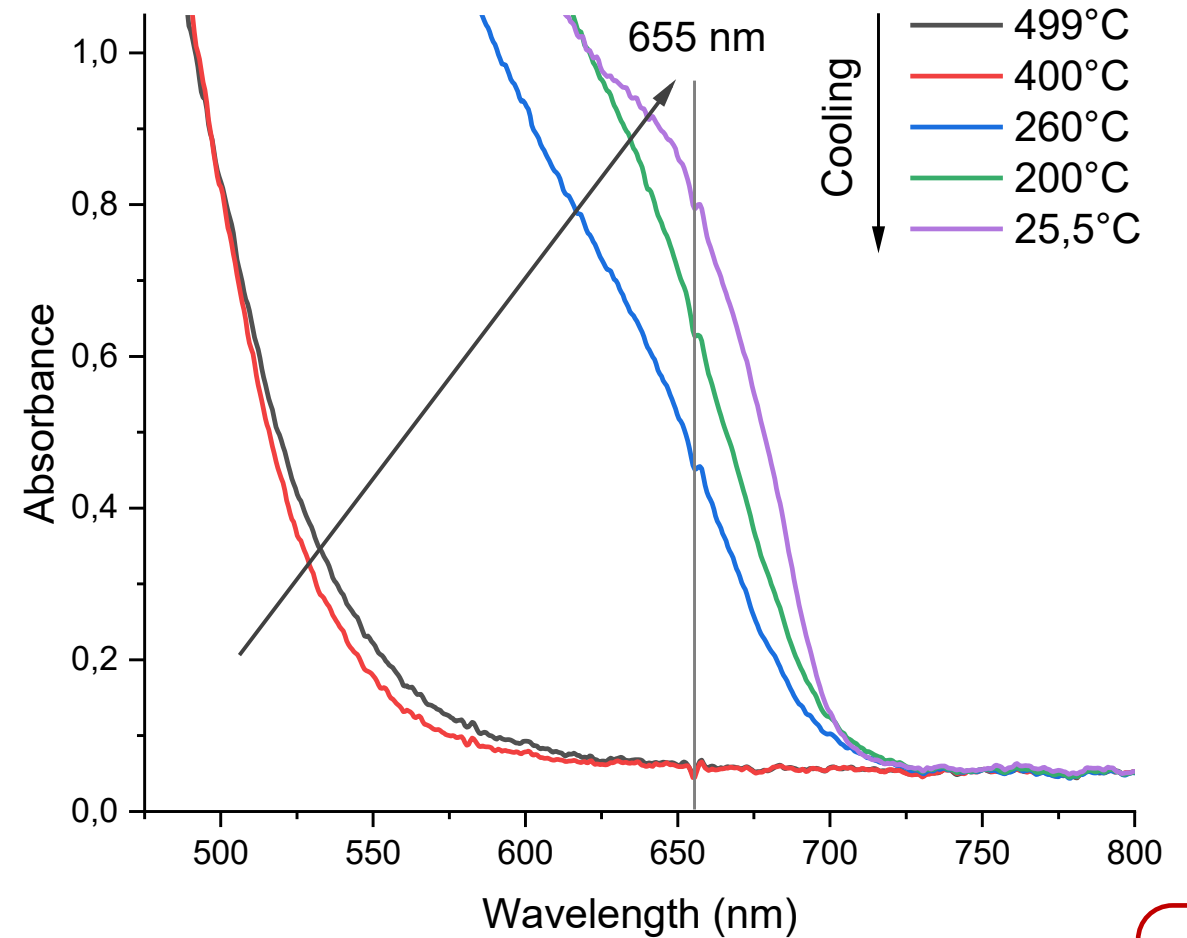
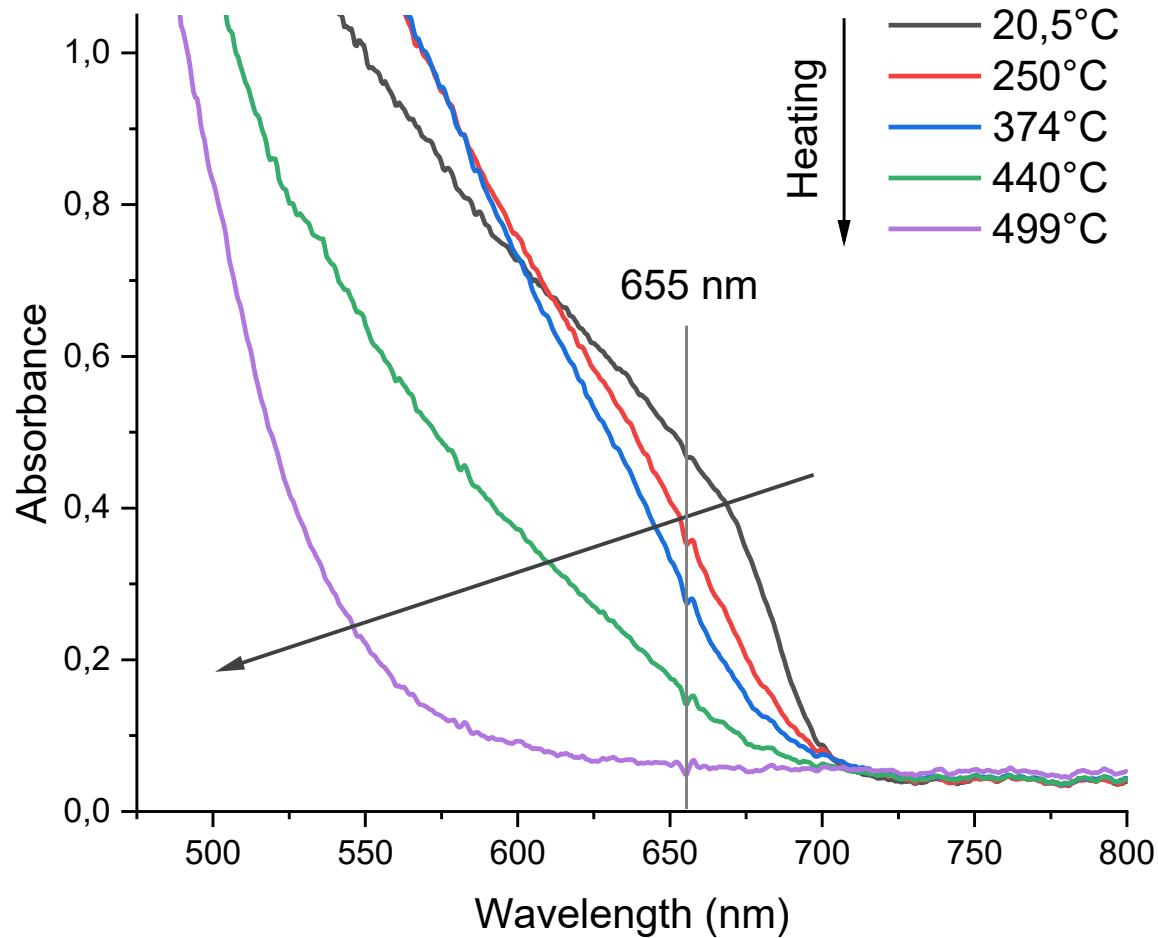
Heating and cooling rates  $2 \pm 0,5$  °C/min

# Absorbance spectra evolution during heating-cooling process

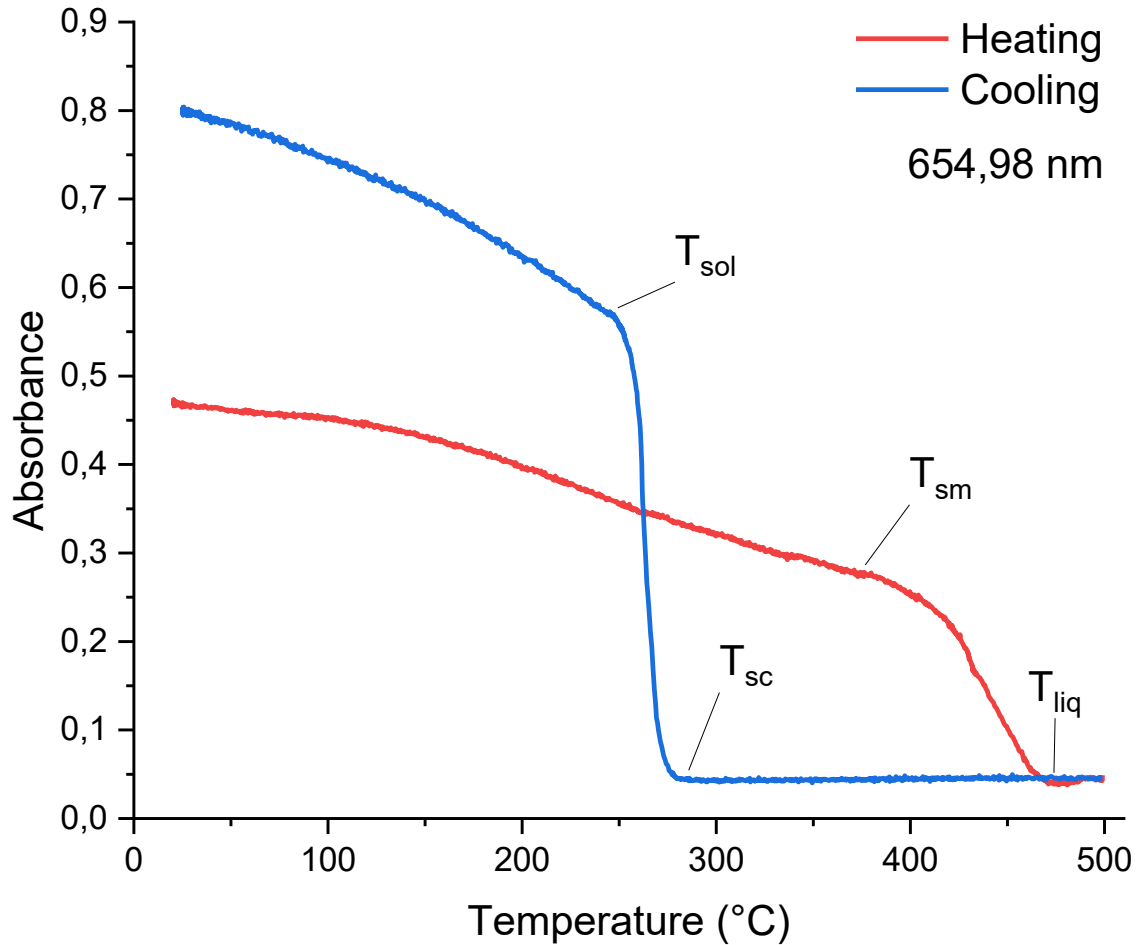


# Absorbance spectra evolution during heating-cooling process

**Br<sub>0</sub>I<sub>1</sub>**



# Dependence of absorbance on temperature



$T_{sm}$  – start of the melting

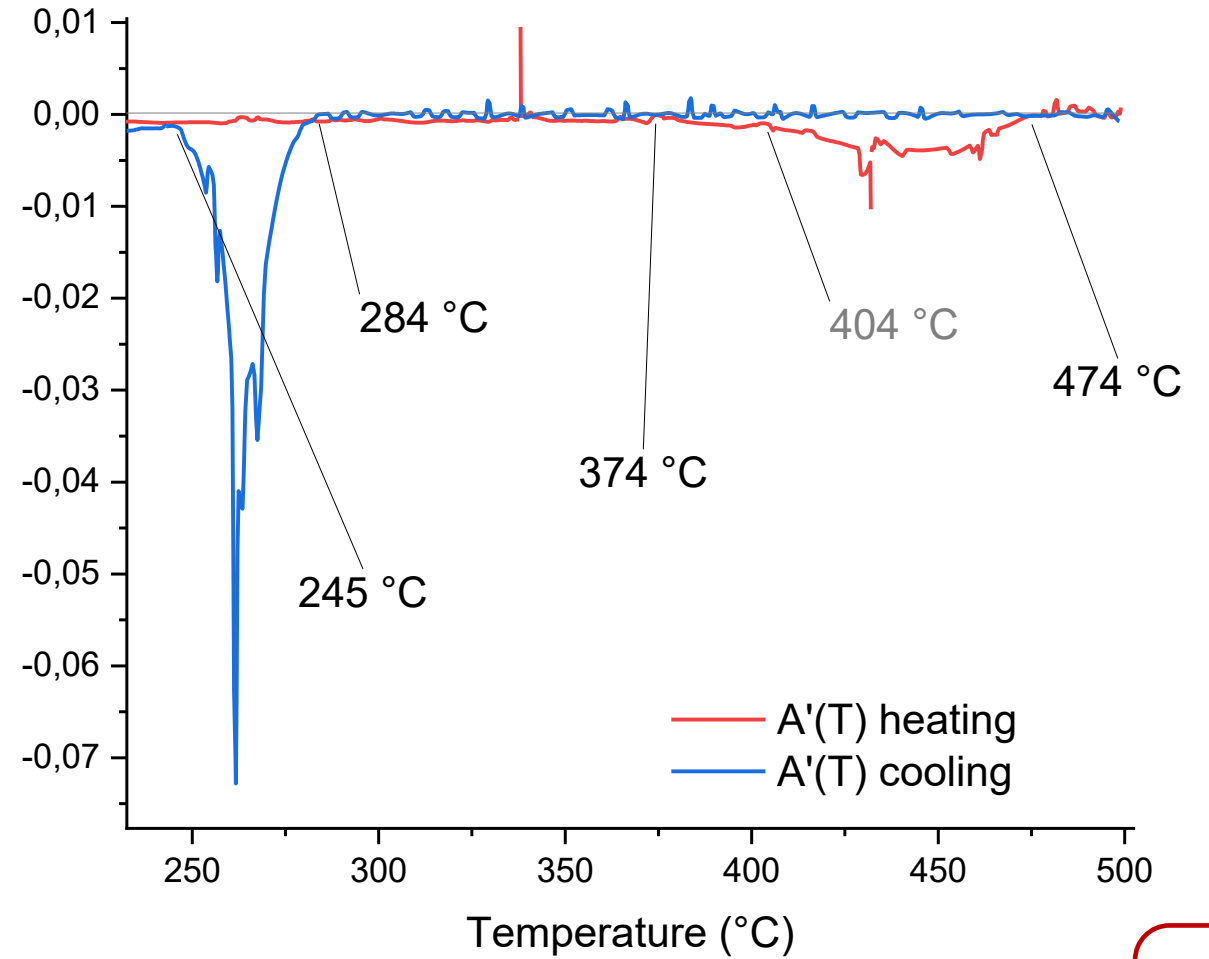
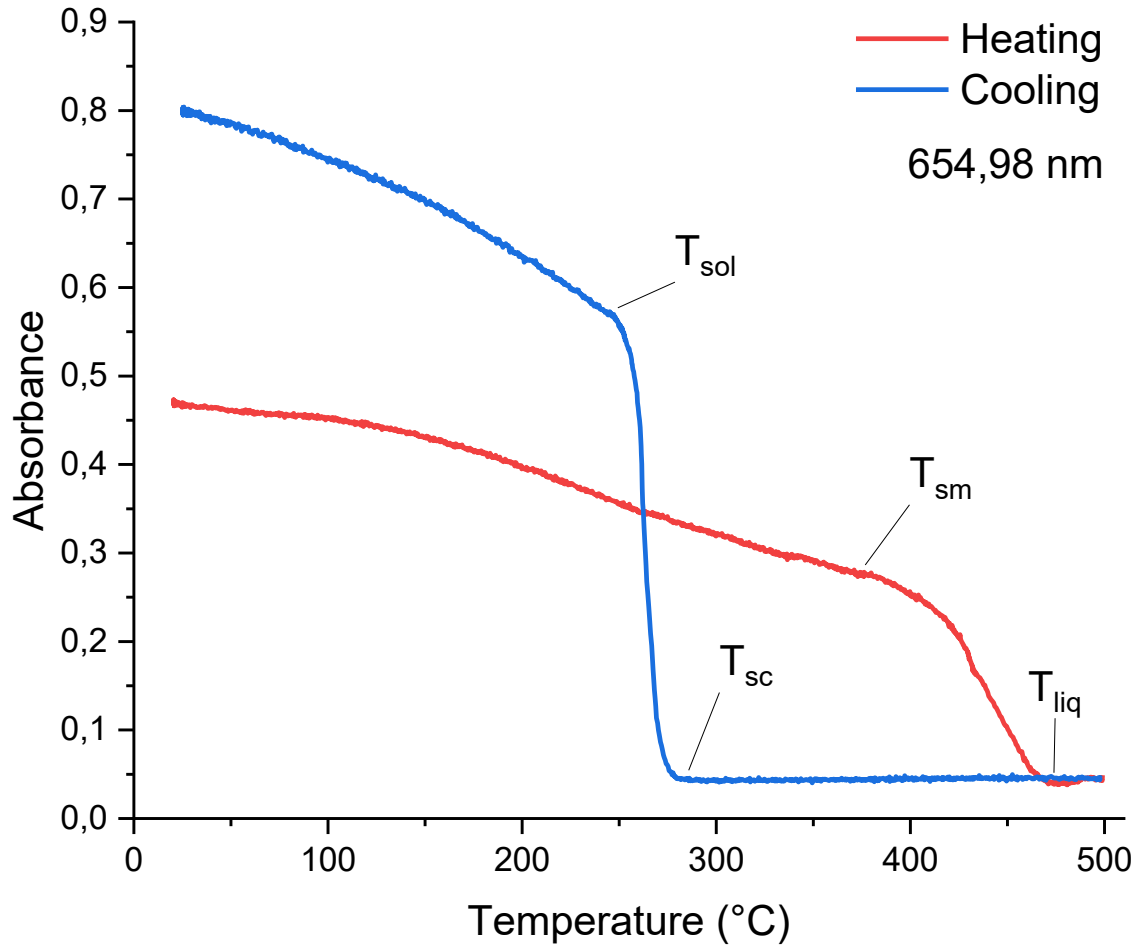
$T_{liq}$  – liquidus

$T_{sc}$  – start of the crystallization

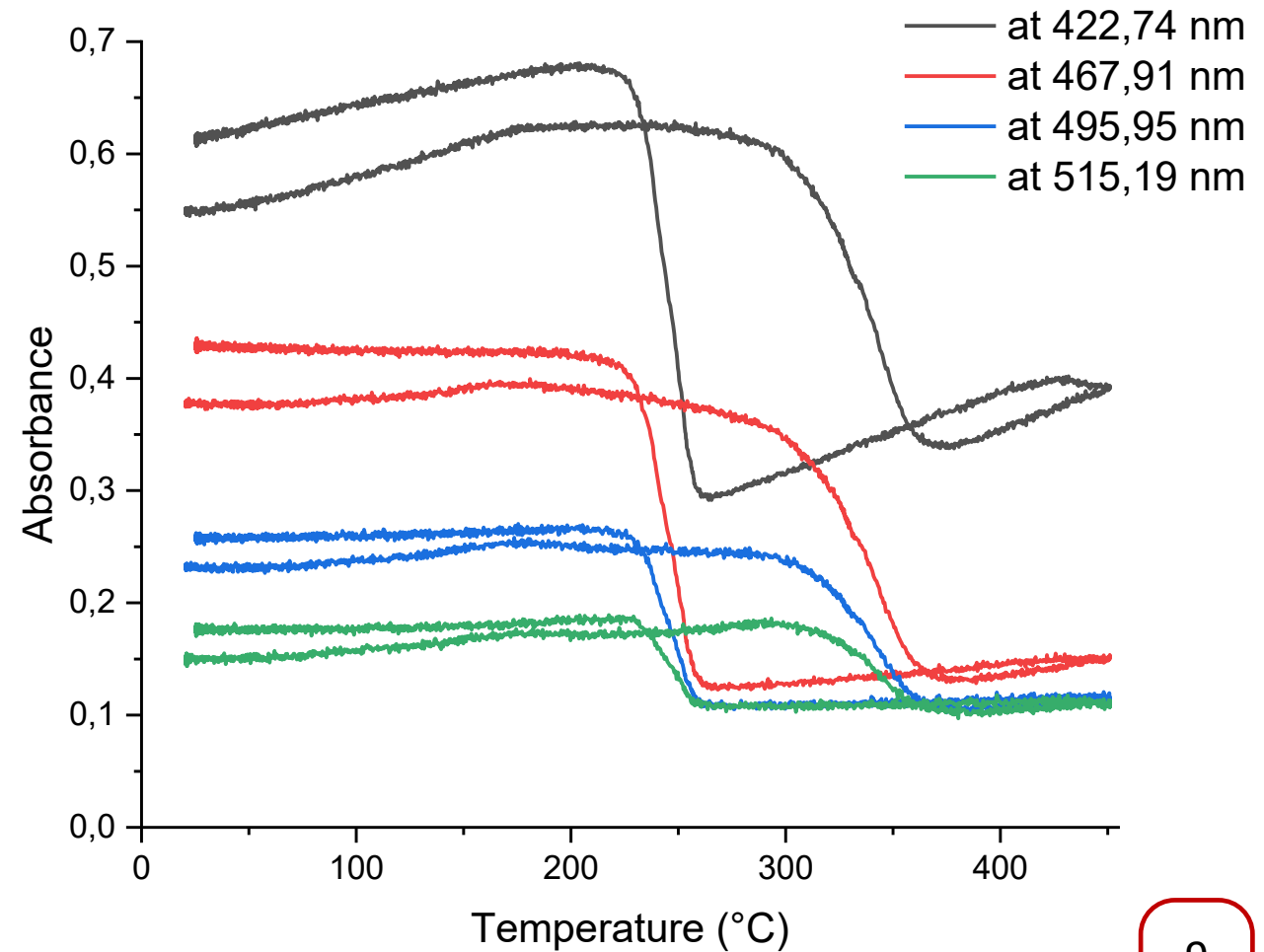
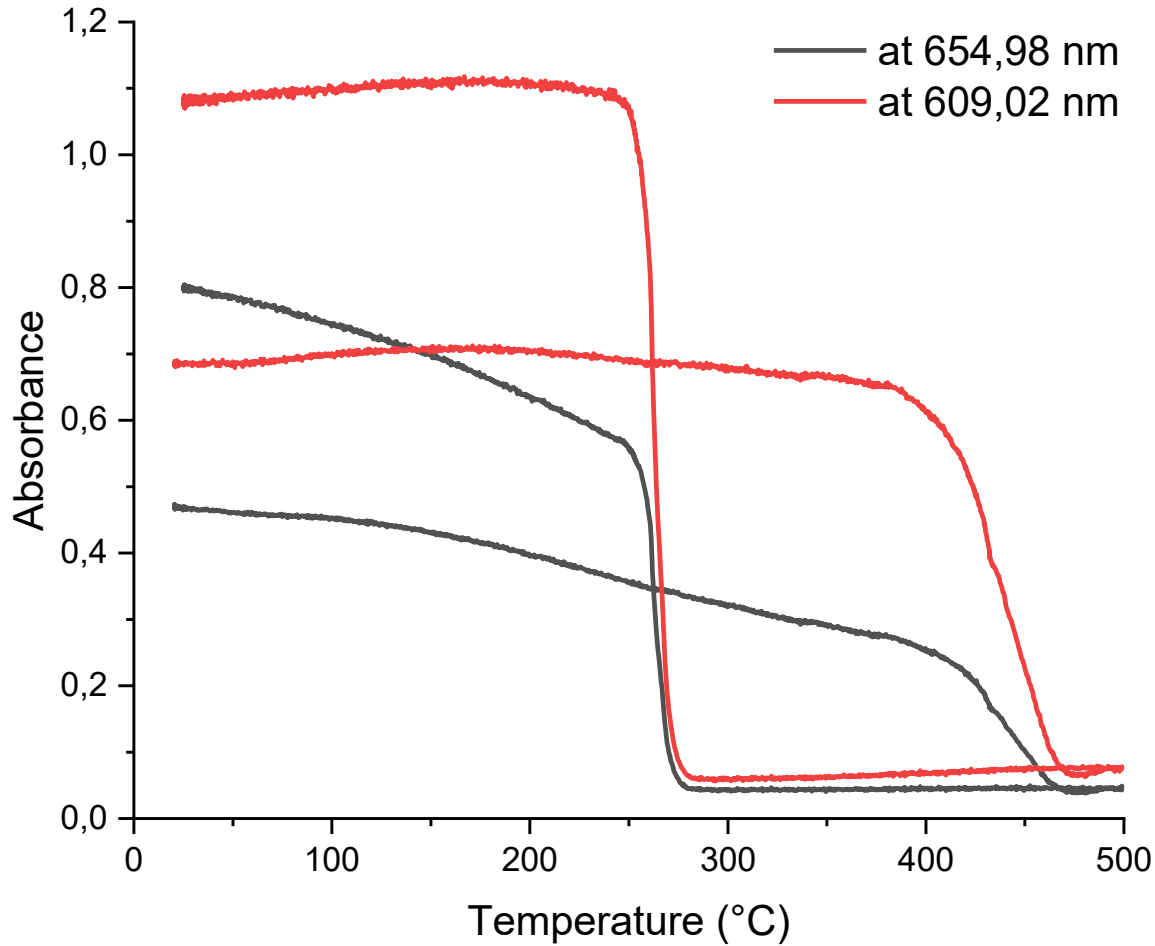
$T_{sol}$  – solidus



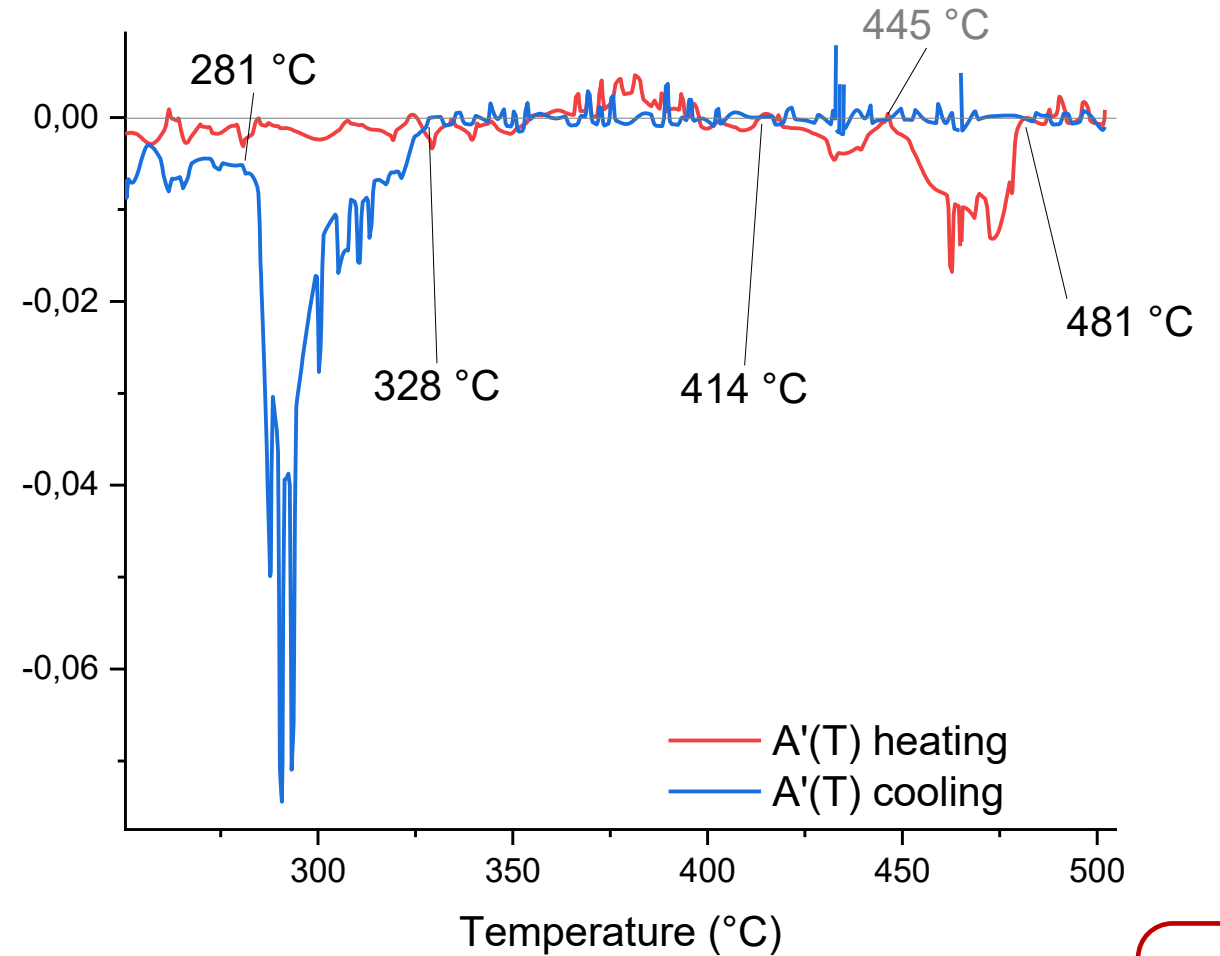
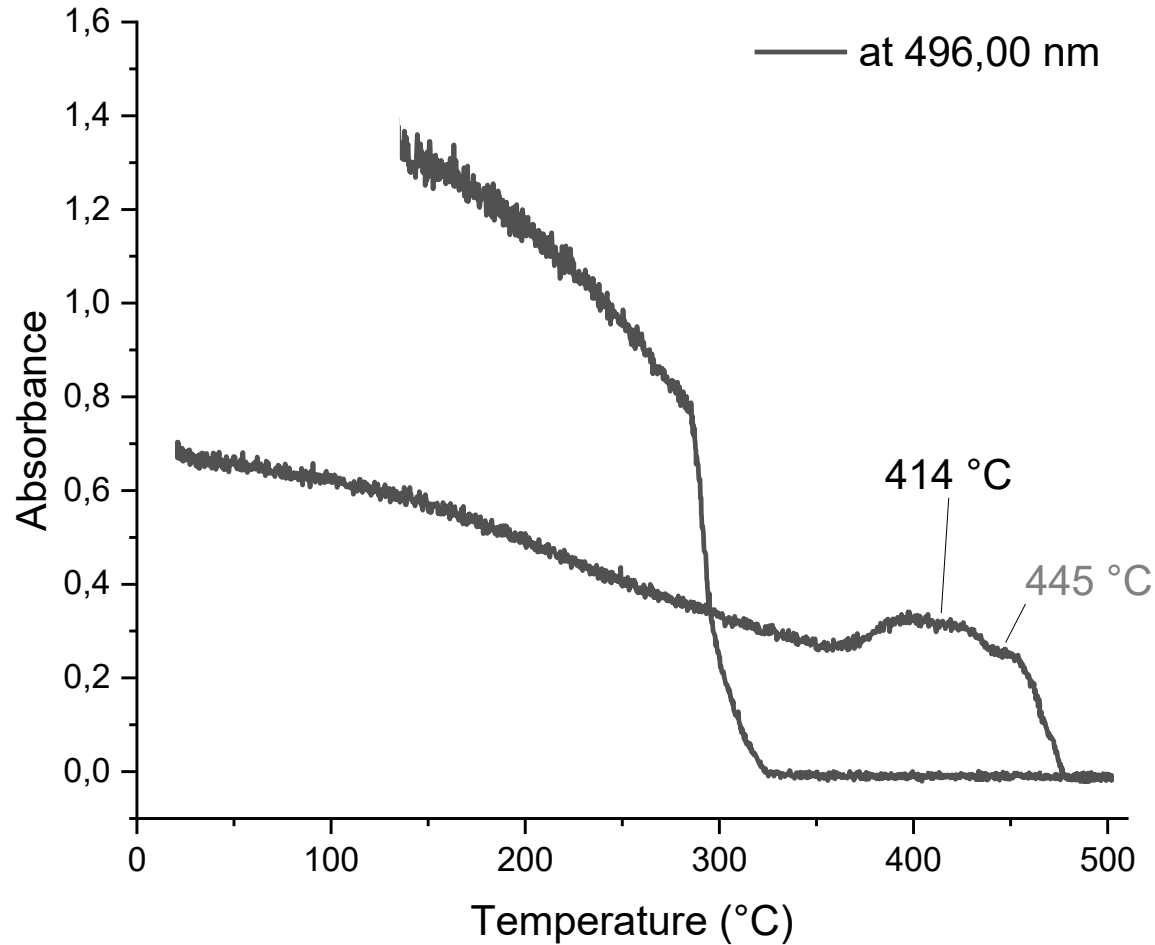
# Dependence of absorbance on temperature



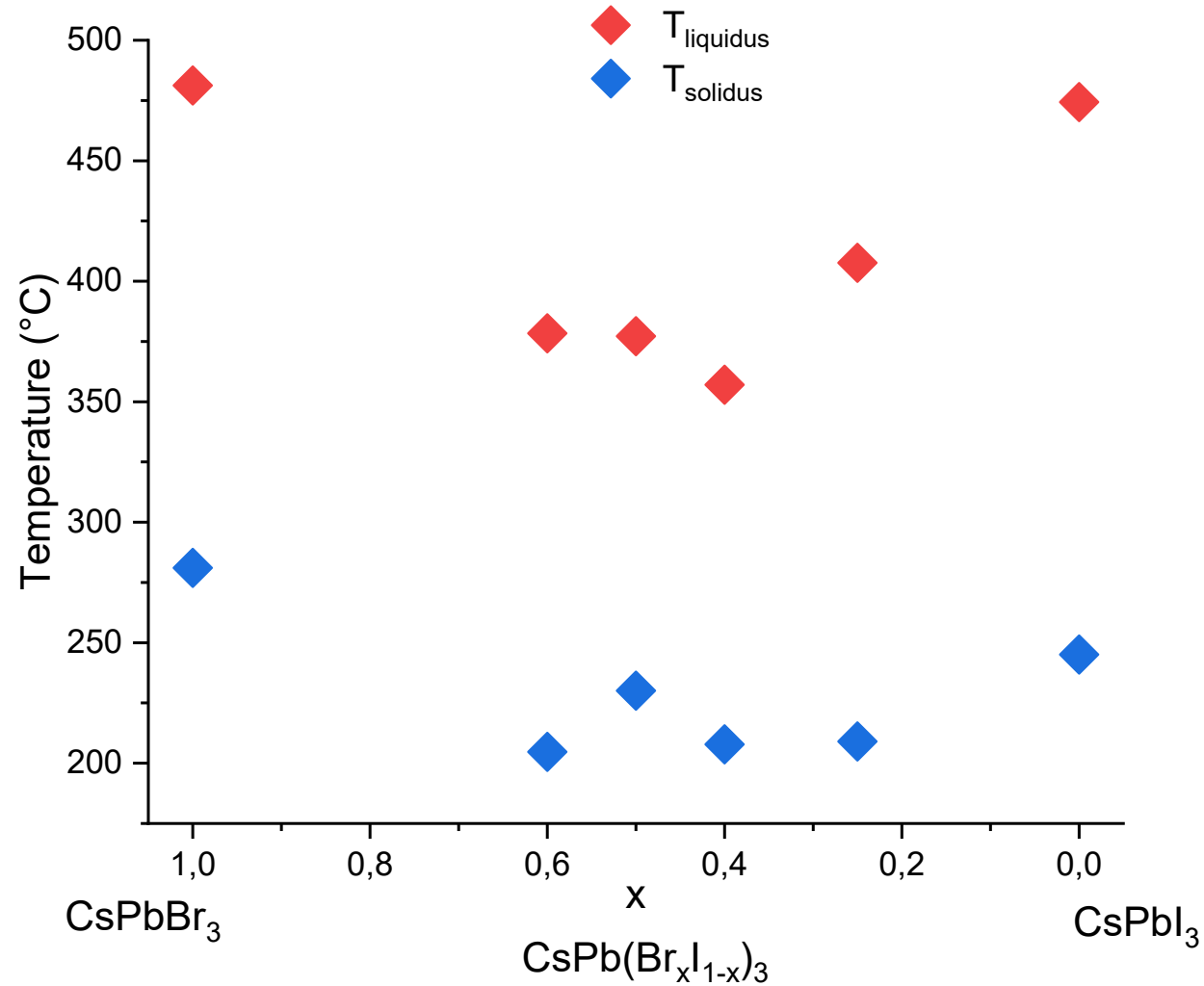
# Dependence of absorbance on temperature



# Dependence of absorbance on temperature



# Dependence of solidus and liquidus temperatures on Br/I ratio



# X-ray induced luminescence

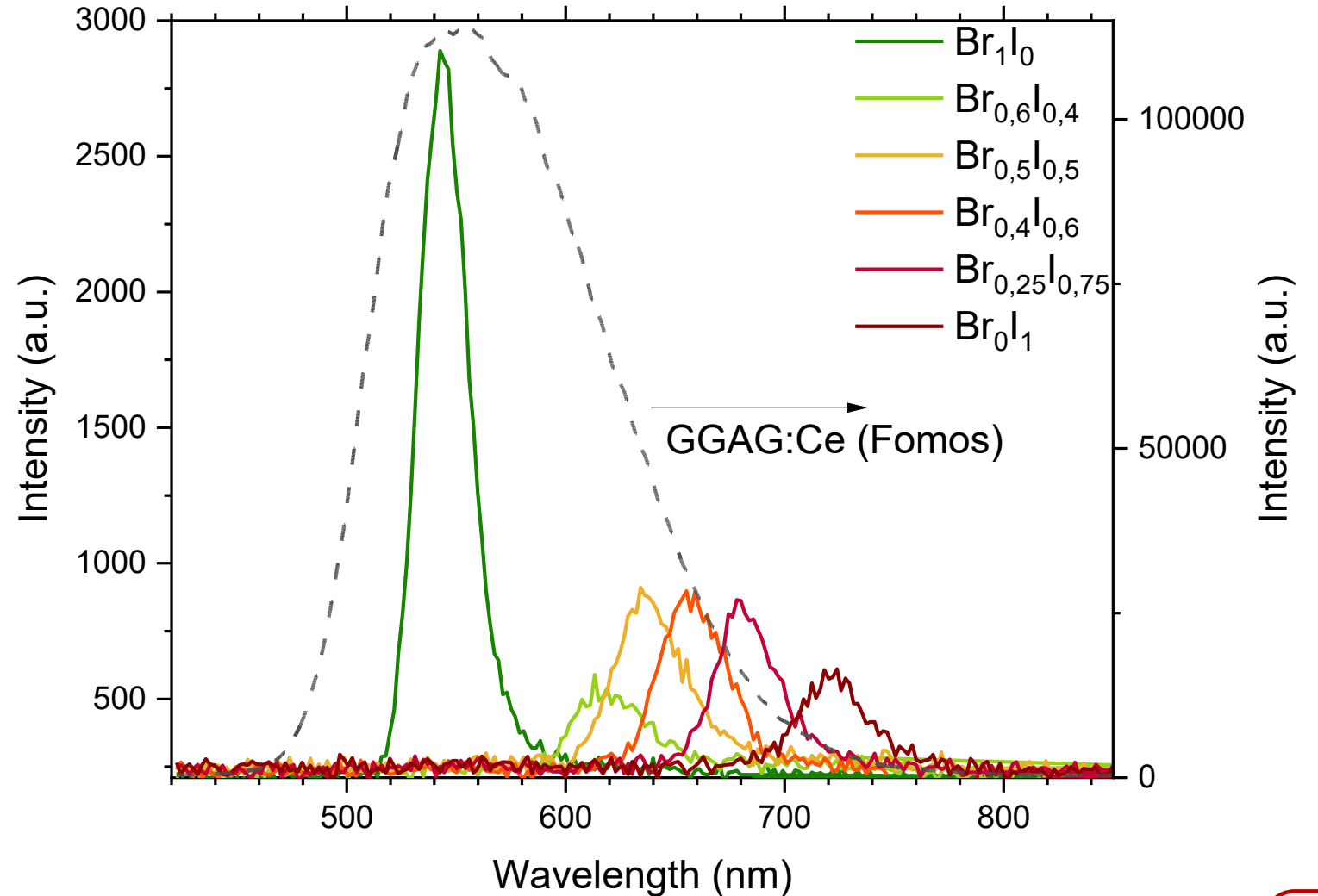
$Br_1I_0$



$Br_{0,6}I_{0,4}$



$Br_{0,5}I_{0,5}$



# Conclusions

- Borogermanate glass-ceramics with mixed  $\text{CsPb}(\text{Br}_x\text{I}_{1-x})_3$  nanocrystals was **synthesized**.
- Equimolar **replacement of bromine with iodine** in glass-ceramics leads to a **shift of absorption spectra** towards **larger wavelengths**.
- **Hysteresis** was observed in the curves of the dependence of the **absorbance** of glass-ceramics **on temperature**.
- The **shapes** of the curves obtained at different wavelengths are **the same**.
- The dependence of the **liquidus** (350 – 500 °C) and **solidus** (200 – 300 °C) **temperatures** on the nanocrystals composition was obtained.
- The synthesized glass-ceramics displays **X-ray induced luminescence**.



Thank you for your  
attention!

harisovarufina@gmail.com



**i**T**MO**re than a  
**UNIVERSITY**