Ab initio study of chemical shifts of X-ray emission spectra in ytterbium halides by the coupled cluster method

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Study of the electronic structure of crystalline compounds

Difficulties:

- High density of low-lying states; (multiconfiguration nature)
- Significant contribution of relativistic effects

=> The need to simultaneously account for relativistic and correlation effects at the highest level of accuracy.

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Description of the crystal fragment:

Density Functional Theory

Exchange & Correlation

Wave Function Theory

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Description of the crystal fragment:



This approach makes it possible:

- description of crystals with impurity centers
- to predict the optical properties of crystals

Step 1: Quantum chemical calculation of the electronic structure

Relativistic coupled cluster method^[1]

[1] Oleynichenko A.V. et al. // Phys. Rev., 2024, V. 109, p. 125106. doi: 10.1103/PhysRevB.109.125106

Step 1: Quantum chemical calculation of the electronic structure

Relativistic coupled cluster method^[1] Generalized relativistic core pseudopotential method^[2] Reducing the number of explicit electrons in Increasing accuracy the calculation

[1] Oleynichenko A.V. et al. // Phys. Rev., 2024, V. 109, p. 125106. doi: 10.1103/PhysRevB.109.125106
 [2] Titov A V, Mosyagin N S. // GRECP: Theoretical grounds, 1999, V. 71, PP. 359-401.

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[2] Titov A V, Mosyagin N S. // GRECP: Theoretical grounds, 1999, V. 71, PP. 359-401.
[3] Lomachuk Y. V. et al. // Phys. Chem. Chem. Phys, 2020, V. 22, PP. 17922-17931. doi: 10.1039/D0CP02277B.
[4] Maltsev D. A. et al. // Phys. Rev. B. 2021. May. V. 103. p. 205105. doi: 10.1103/PhysRevB.103.205105.
[5] Shakhova V. M. et al. // Phys. Chem. Chem. Phys. 2022. V. 24. PP. 19333–19345. doi: 10.1039/D2CP01738E.
[6] Oleynichenko A.V. et al. // Phys. Rev., 2024, V. 109, p. 125106. doi: 10.1103/PhysRevB.109.125106.

Step 2: Restoration method^[7]



Obtaining a four-component wave function after calculation with the pseudopotential

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Chemical shifts of X-ray emission spectrum:

Pic 3: Diagram of the K-, L- and M-levels of atomic energy





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Chemical shifts of X-ray emission spectrum:

Obtaining a four-component wave function after calculation with the pseudopotential

Step 3: Method for calculating chemical shifts of X-ray emission spectrum lines^[8]

Calculation of the difference quantity upon excitation of inner core states



Pic 3: Diagram of the K-, L- and M-levels of atomic energy



Research objects:

Crystal fragments with CTEP for YbHal_n(Hal=F, Cl; n=2, 3)

Method of calculations:	Program:	↓ <mark>9 Basis set:</mark> Vb:	F — 4s3p1d Cl — 5s4p1d
Relativistic Coupled Clusters (RCC-SD)	EXP-T	4s4p2d2f for compound with Yb ²⁺ 6s6p4d4f for compounds with Yb ³	





Pilot calculations of chemical shifts of X-ray emission spectrum lines in crystal fragment with CTEP

СТЕР	κ _{α1}	κ _{α2}		
YbF3/YbF2	513	452		
E	579±26 ^[9]	570±114 ^[9]		
Experiment	557±27 ^[10]			
YbCl3/YbCl2	517	455		
Experiment	574±35 ^[11]			



Table 1: Chemical shifts of the XES lines of the Yb atom in the YbHal₃ crystal relative to the YbHal₂ crystal, meV

[8] Matsushita T., Hofmann H. F. Origin of meter fluctuations in weak measurement interactions // Phys. Rev. A. 2024. V. 109. p. 022224.
 [9] E.V. Petrovich et al. Valence states of rare earth elements according to data on chemical displacements of X-ray lines // Radiochemistry. 1976. № 288
 [10] A. Sovestnov, Academy of Sciences of the USSR, Leningrad Institute of Nuclear Physics named after B.P. Konstantinov, 1982

Contributions to the corrections to the mc-PP-CCSD calculations





$$K_{average} = \frac{1}{3} (2K_{\alpha 1} - K_{\alpha 2})$$

Table 2: Chemical shifts value for ytterbium fluorides, meV

YbF3/YbF2	Molecules			Crystal fragment with CTEP		Eventiment
	mc-PP	sc-PP	Δ	mc-PP	mc-PP+∆	Experiment
κ _{α1}	539	630	92	513	605	579±26 ^[8]
κ _{α2}	473	559	85	452	538	570±114 ^[8]
Kaverage	517	607	90	493	583	557±27 ^[9]

Table 3: Chemical shifts value for ytterbium chlorides, meV

	Molecules		Crystal fragment with CTEP		Eventiment	
	mc-PP	sc-PP	Δ	mc-PP	mc-PP+Δ	Experiment
κ _{α1}	521	605	84	517	601	
κ _{α2}	457	536	79	455	534	
Kaverage	500	582	82	496	579	574±35 ^[10]



For the first time, the chemical shifts of the X-ray emission spectrum on heavy atoms in crystals were calculated using the relativistic coupled cluster method

From the results obtained it is clear that in order to calculate the properties near a heavy atom it is necessary to carry out high-precision calculations of the electronic structure

Thanks for your attention