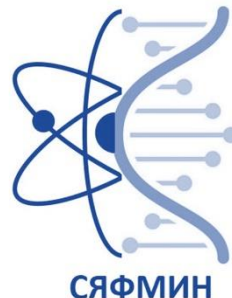


A study of the defect formation of CuO nanostructures under He⁺ ion irradiation

S. Fortuné Fábregas

silviafortune@jinr.ru

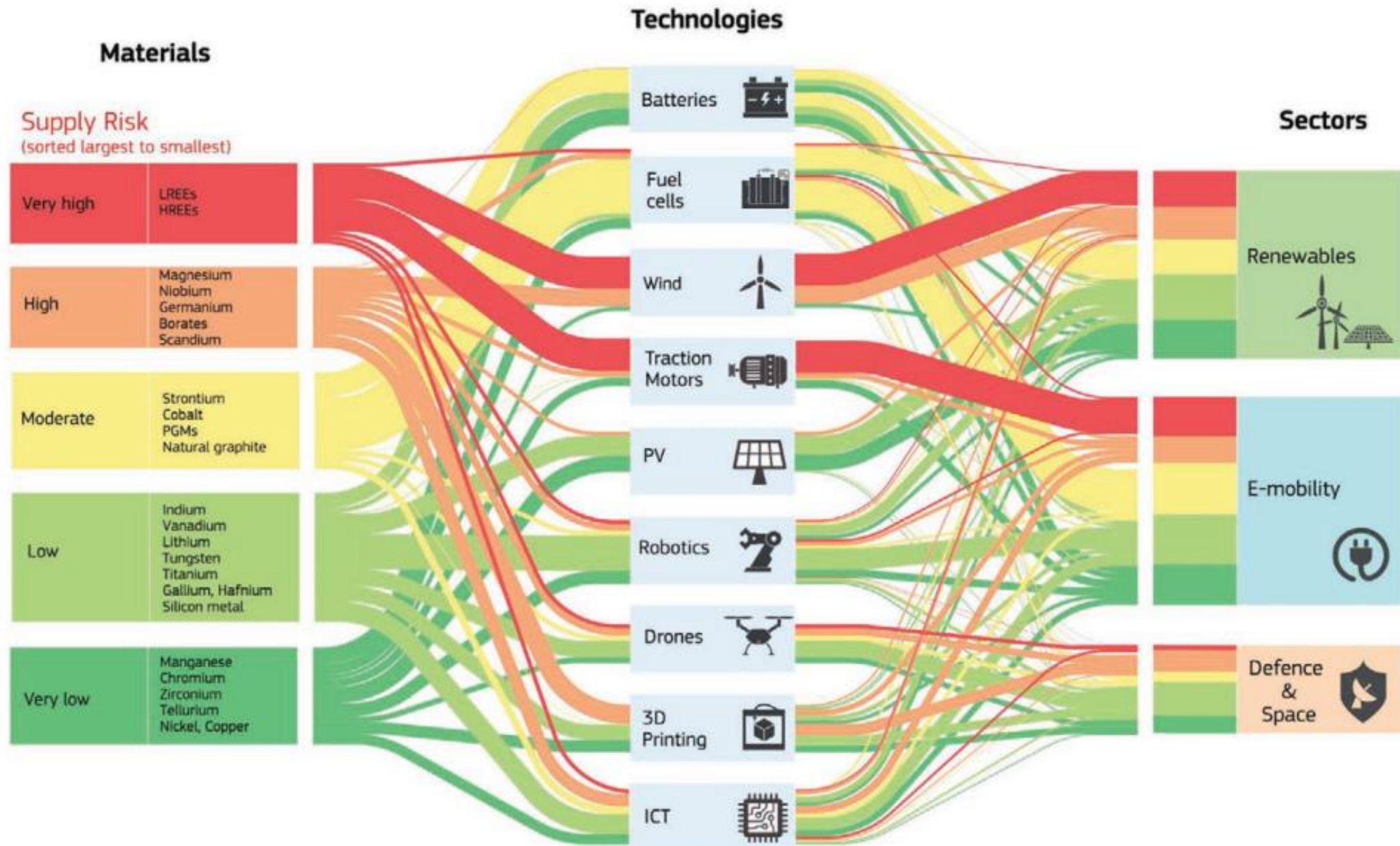
- *Photovoltaic Laboratory, Institute of Materials Science and Technology of the University of Havana (IMRE-UH), Vedado, 10 400 Plaza de La Revolución, Havana, Cuba.*
- *Sector of Nuclear-Physical Materials Science and Ion-Implantation Nanotechnologies (SNPMSIIN), Frank Laboratory of Neutron Physics, (FLNP), Joint Institute for Nuclear Research (JINR), 141 980 Dubna, Moscow oblast, Russia.*



Content

1. Introduction
 - a) CuO properties
2. Experimental technique
 - a) CuO nanostructures synthesis
 - b) CuO nanostructures irradiation
3. Results
 - a) CuO spin coating. SEM, Raman and XRD.
 - b) CuO irradiated. XRD, DBS and PALS.
4. Conclusions

1. Introduction. CuO properties.



1. Introduction. CuO properties.

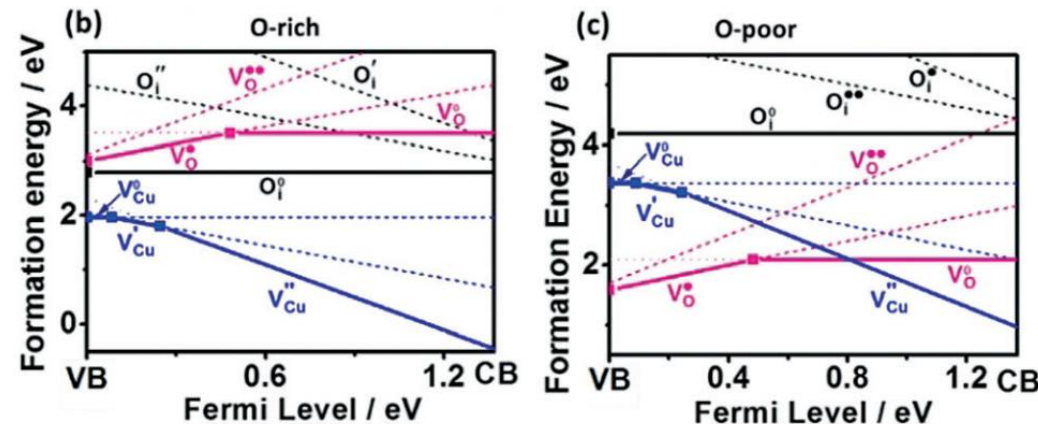
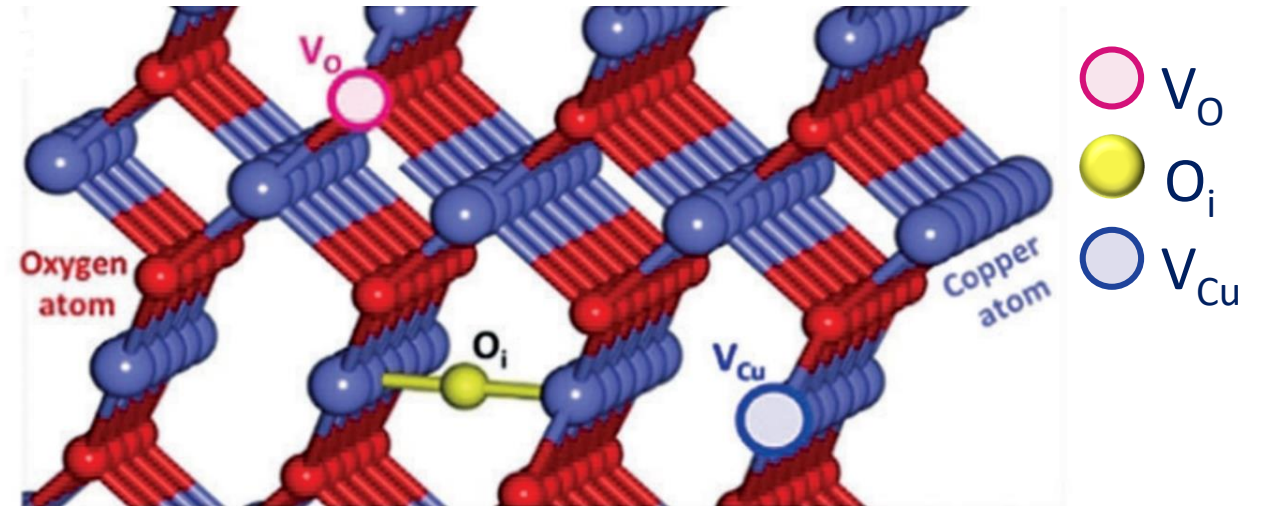
Properties:

- Non-toxic and abundant in the Earth's crust.
- Easy to obtain by low-cost techniques.
- p-type semiconductor with a narrow bandgap of 1.2-2.16 eV.
- Low hole mobility $0.1-10 \frac{cm^2}{V \cdot s}$
- Electronic conductivity $1.05 \cdot 10^{-2}$ and $6.30 \cdot 10^{-2} (\Omega \cdot cm)^{-1}$.
- The dominant defects in CuO are the copper vacancies (V_{Cu}).

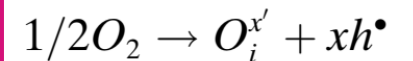
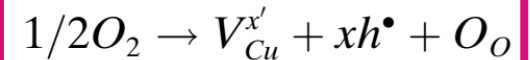
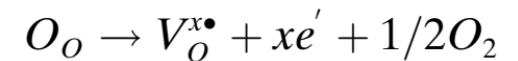
How to modify CuO properties?

- thermal treatments,
- doping,
- nanocompounds,
- irradiation.

Types of defects in CuO

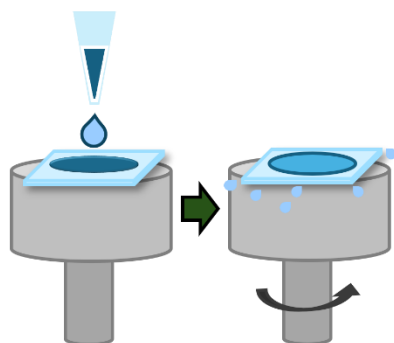


Defect reactions:

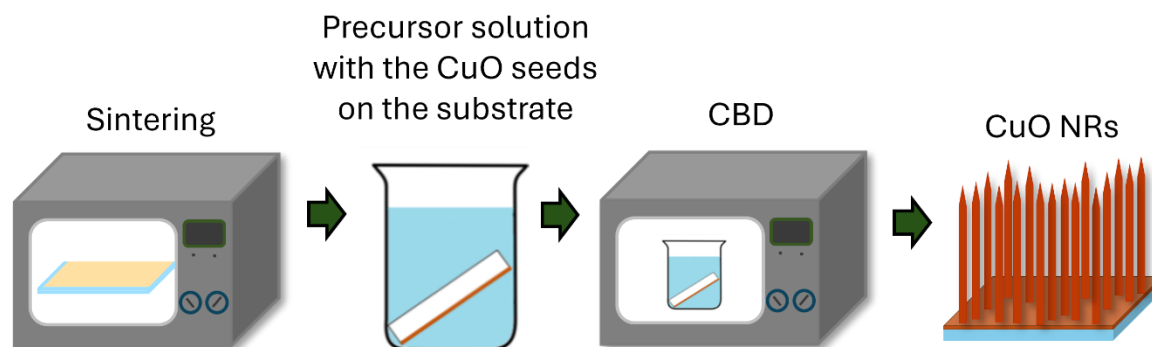


2. CuO nanostructures synthesis and irradiation

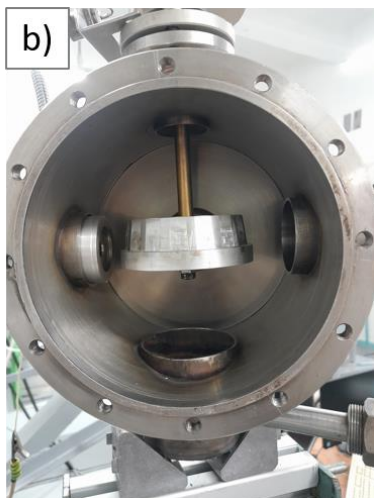
CuO seeding process



CuO elongated nanostructures synthesis



Sample holder and irradiation

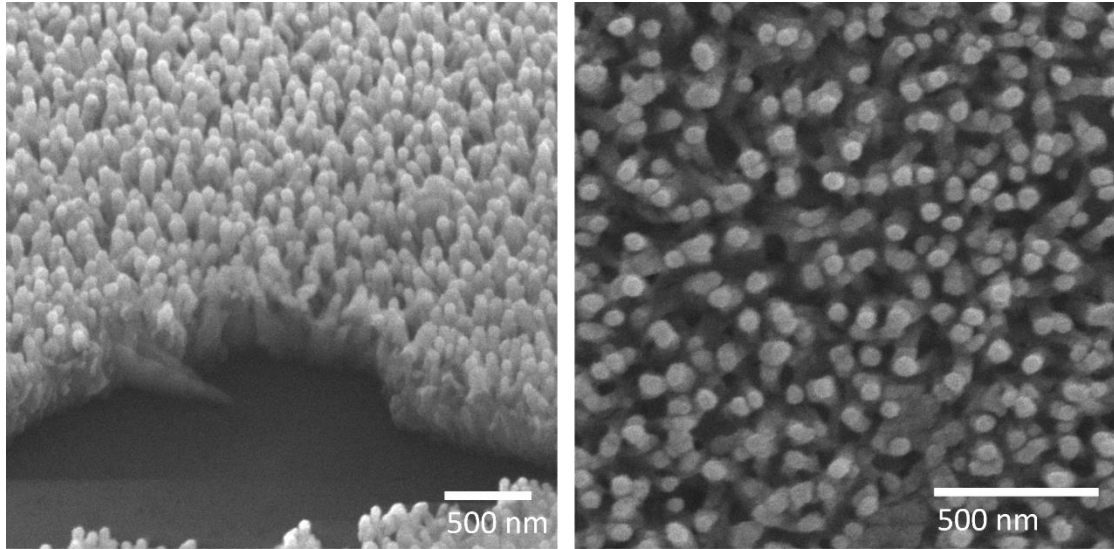


CuO irradiated samples parameters

Sample	Dosis (part/cm ²)	time (s)
CuO reference	0	0
CuO_10	$3.13 \cdot 10^{13}$	10
CuO_100	$3.13 \cdot 10^{14}$	100
CuO_1000	$3.13 \cdot 10^{15}$	1000
CuO_5000	$3.13 \cdot 10^{16}$	5000
CuO_3h	$3.38 \cdot 10^{16}$	10 800

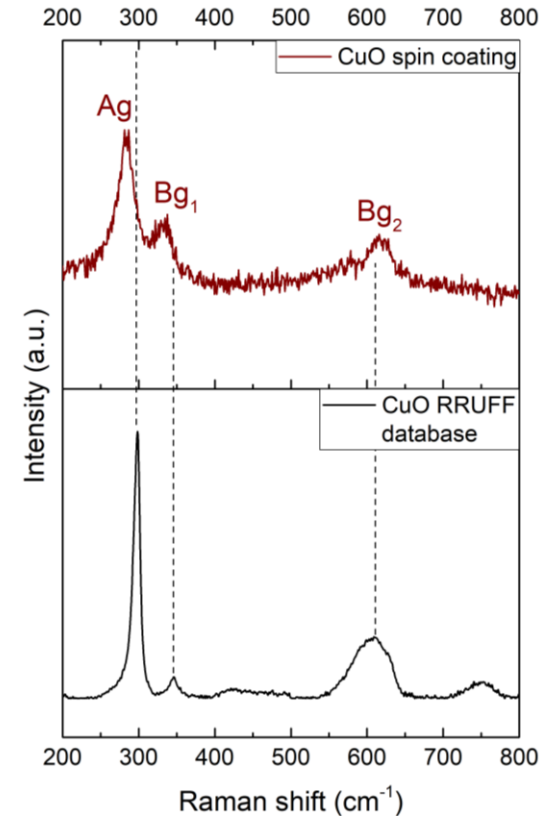
3.1. CuO nanostructures

SEM

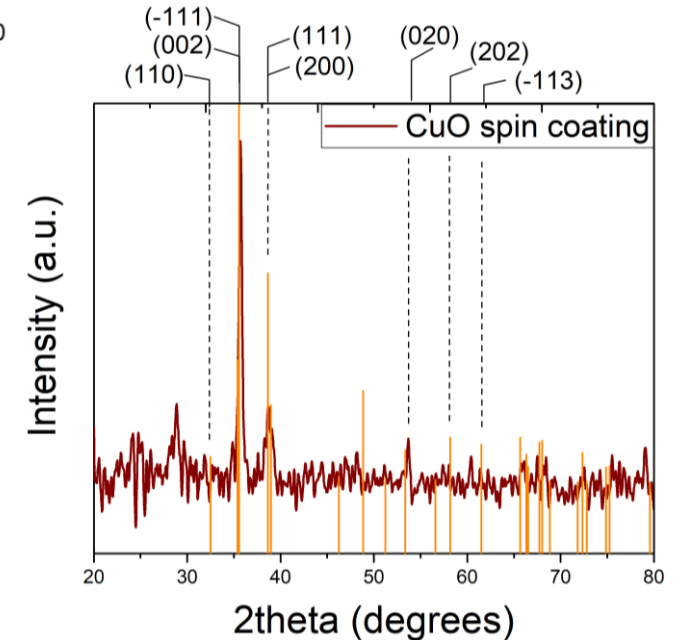


- CuO structures with elongated morphology.
- CuO nanostructures diameter was calculated to be (64 ± 1) nm.
- CuO nanostructures height was estimated to be about 430 nm.

Raman spectroscopy



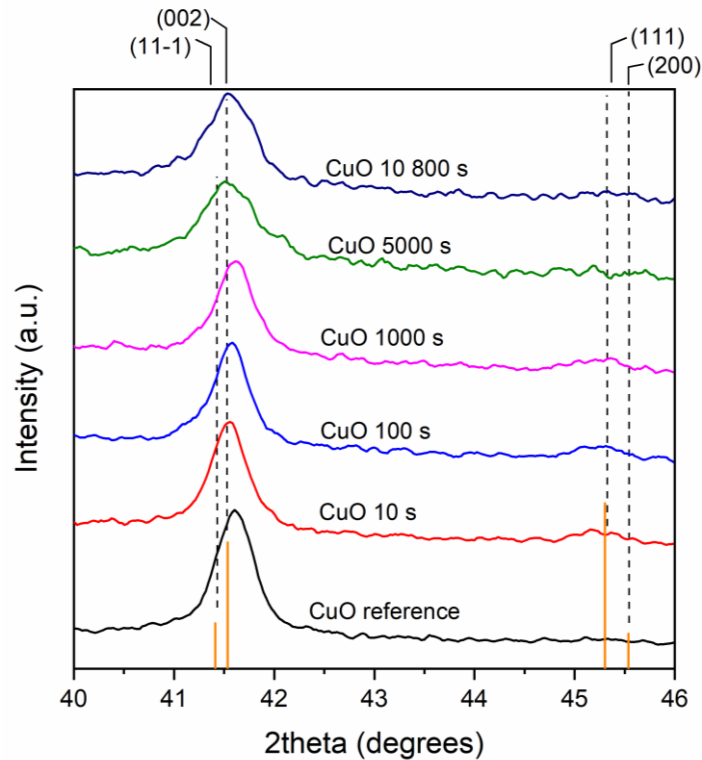
XRD



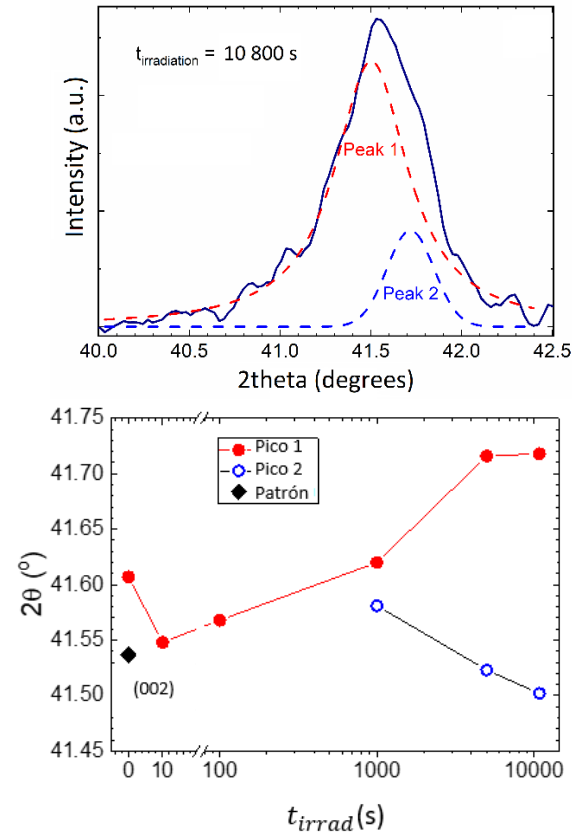
- Raman spectroscopy and XRD confirm the presence of CuO with a monoclinic structure in the samples.
- The XRD pattern shows a preferential orientation in the direction [001].

3.2.1. CuO irradiated nanostructures: XRD

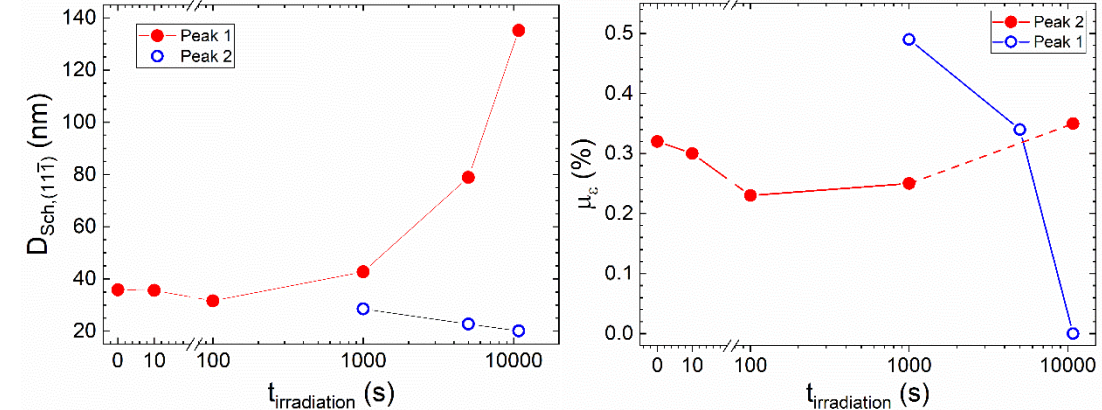
XRD diffractograms of the irradiated samples



Deconvoluted diffraction maxima (11-1)(002)



Crystallite size and strain



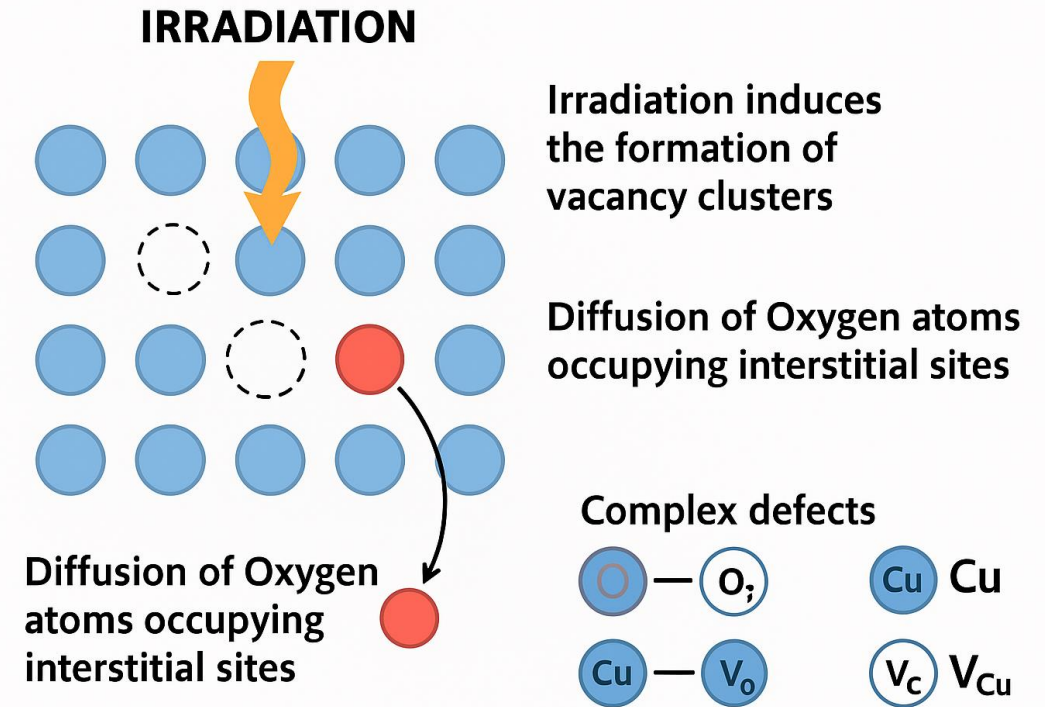
- Increase in dosis produce bigger crystallites and more strain.

- 2theta decreases at $t_{\text{irrad}} < 10\text{ s}$ dosis.
- 2theta increases at higher dosis.
- Cristalline lattice under tensión.
- 2nd peak shift to low 2theta at higher dosis: cristal lattice locally tenses.

3.2.1. CuO Irradiated nanostructures: XRD

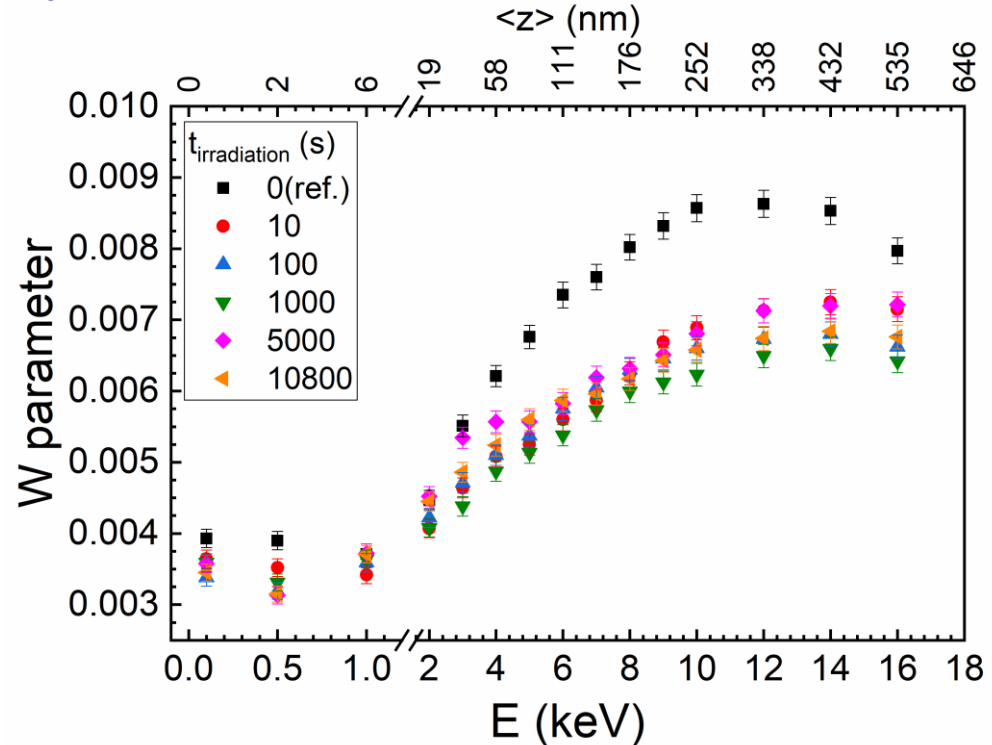
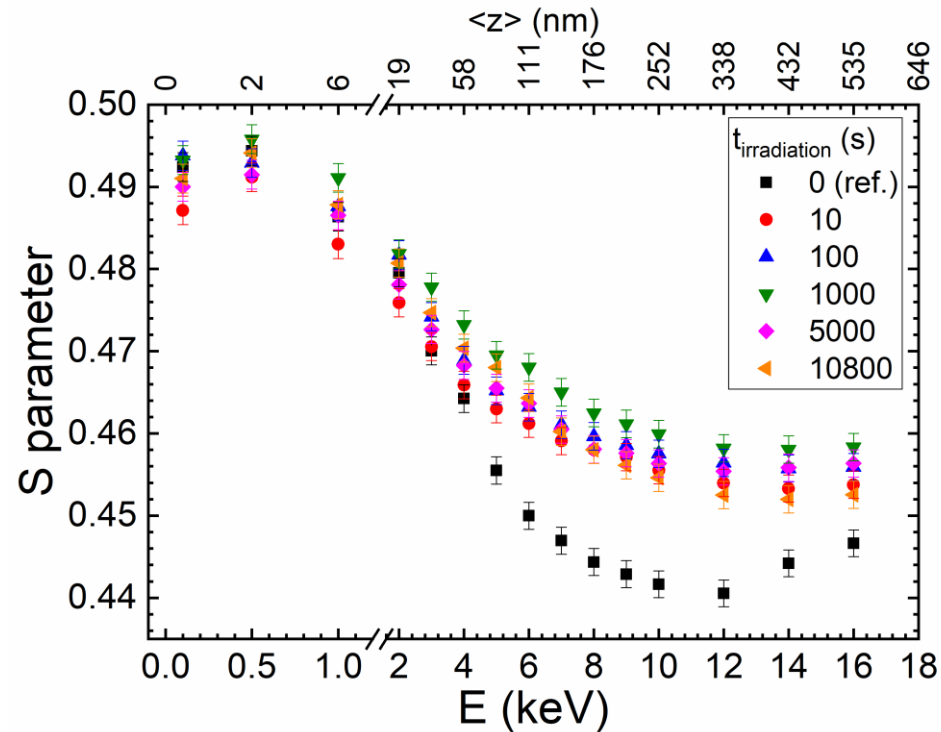
Influence of the irradiation in the lattice:

1. formation of vacancy clusters.
2. diffusion of Oxygen atoms from their original positions in the lattice, occupying interstitial sites (O_i).
3. complex defects such as $\text{Cu}-V_O$ or $\text{Cu}-V_{\text{Cu}}$.



3.2.2. CuO Irradiated nanostructures: DBS

S and W parameters



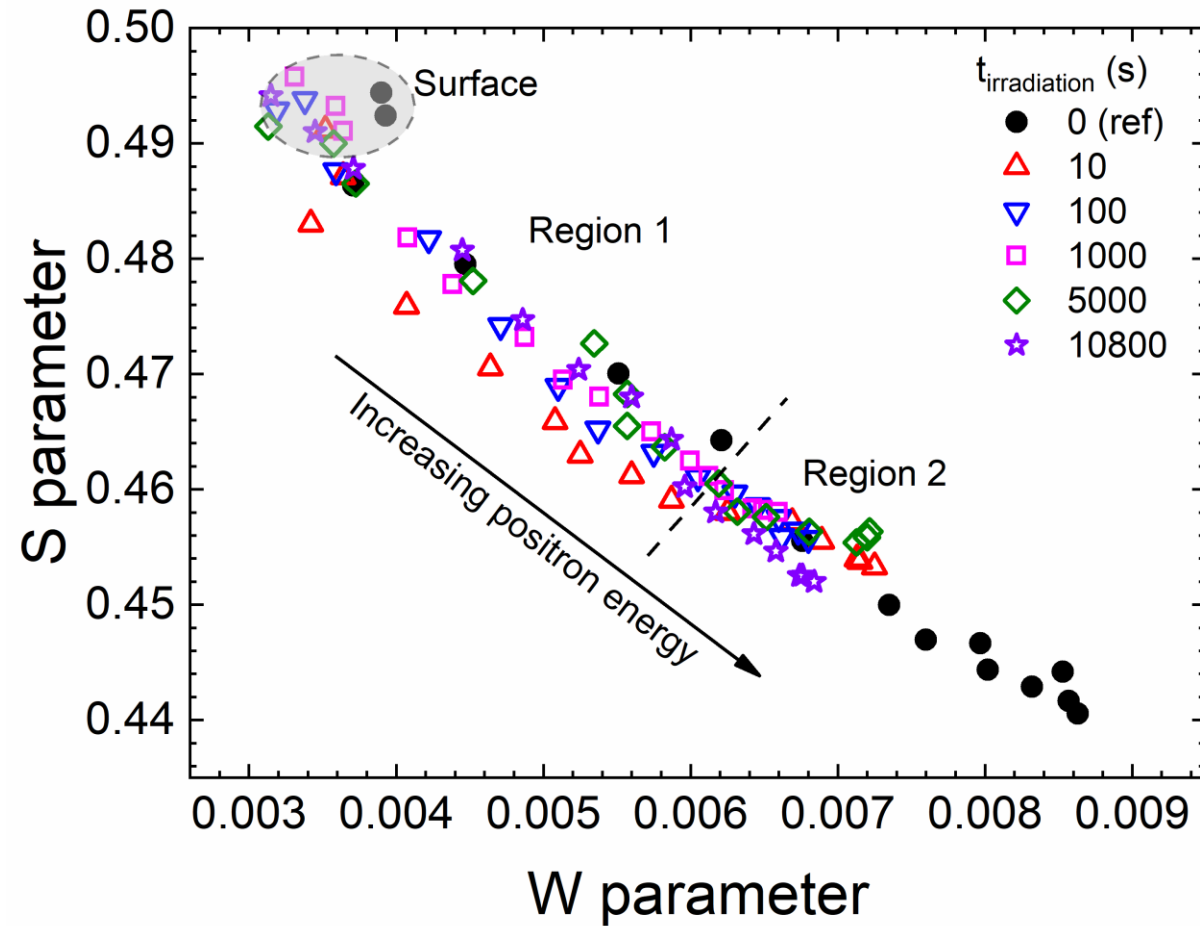
- The increase in **S** parameter indicates that the number of monovacancies (V_{Cu} , V_O), in the material increases.
- Low doses: defects are generated by irradiation.
- High doses: punctual defects form clusters.



Competition between:
Irradiation and the CuO lattice recovery.

3.2.2. CuO Irradiated nanostructures: DBS

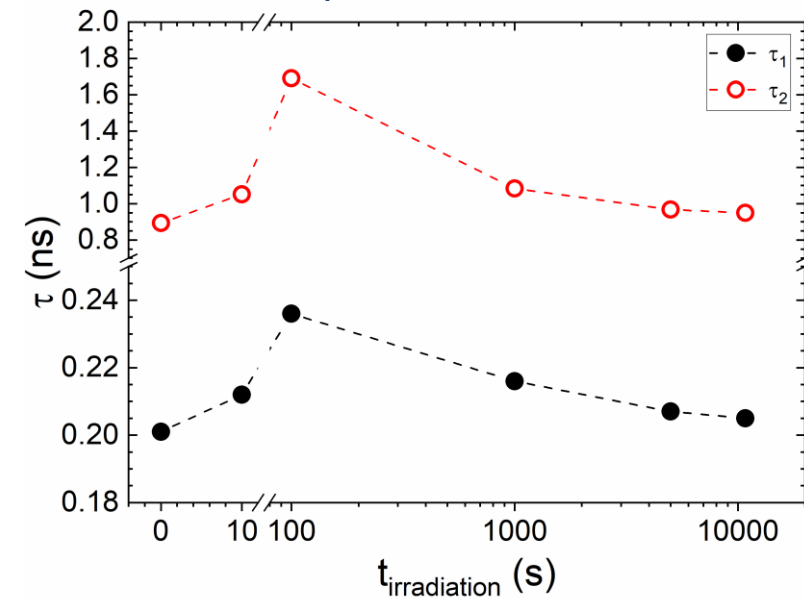
Correlation between S and W



The inequality between regions 1 and 2 is due to the considerable difference between their slopes (trapping fractions).

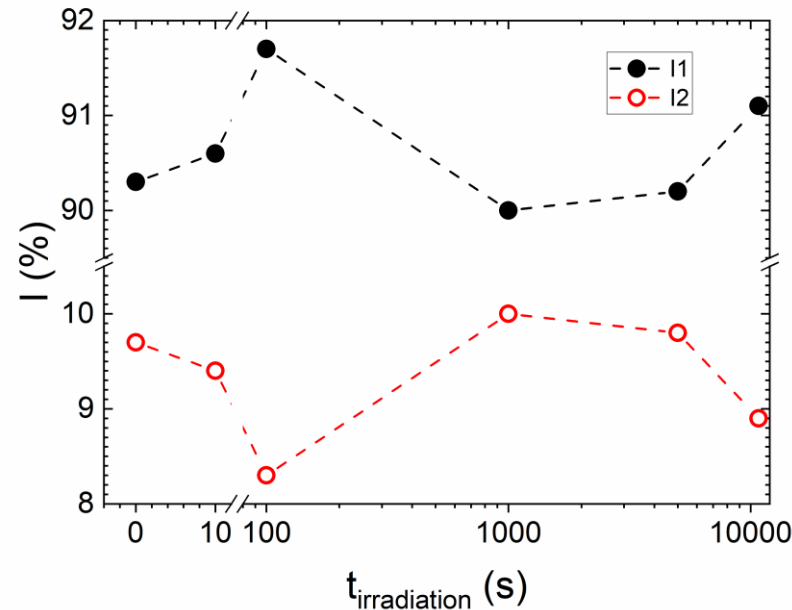
3.2.3. CuO Irradiated nanostructures: PALS

CuO positron lifetimes



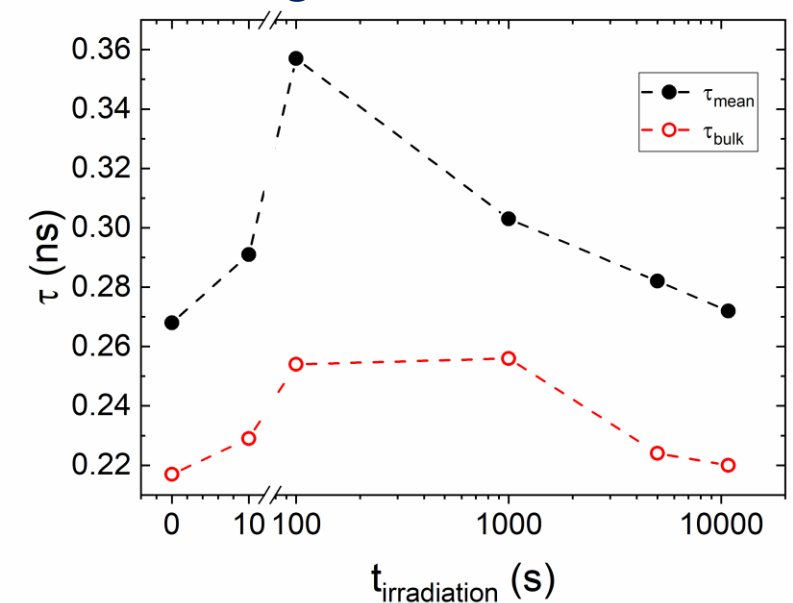
- These lifetimes correspond to V_{Cu} .
- At higher t_{irrad} τ decreases until 0.205 ns.
- $\tau_2 \geq 0.893$ ns are correlated with vacancy clusters present on the surfaces of the nanostructures.

Intensities



- $I_1 \rightarrow 90\text{-}91.7\%$: V_{Cu} **predominant defect**.
- I_1 shows similar behavior to τ_1 , except in CuO 3h. This indicates that the defect concentration does not increase.
- defects associated to I_2 large in size but low in concentration.

Average and bulk lifetimes



- τ_{ave} and τ_{bulk} exhibit a similar behavior to τ_1 y τ_2 .
- τ_{ave} and τ_{bulk} decreases about a 35% at **higher dosis due to a decrease of superficial and bulk vacancies**.

3.2.3. CuO Irradiated nanostructures: PAS

Influence of the irradiation in the lattice:

1. Increase of monovacancies (V_{Cu} , V_O), with the dosis.
2. Formation of punctual defect clusters.
3. V_{Cu} predominant defect.
4. Higher dosis: decrease of superficial and bulk vacancies.



CuO nanostructures
conductivity increase!!

4. Conclusions

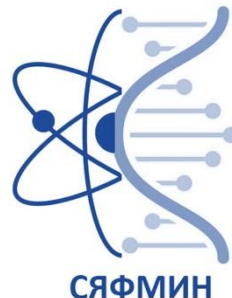
This work exposed the results of the structural and morphological analysis and the evolution of defects in irradiated copper (II) oxide nanostructures. For this purpose, CuO was subjected to helium ion irradiation treatment. This process led to novel results regarding the influence of irradiation on CuO nanostructures. Specifically, concerning to structural variations, since the irradiated samples show novel results because the defects induced by irradiation compete with the recovery of the crystal lattice. Besides, this irradiation produce a large amount of V_{Cu} the type of defect that is predominant and produce holes, increasing the CuO conductivity. This improvement contributes to produce better CuO nanostructured layers for optoelectronics applications.

A study of the defect formation of CuO nanostructures under He⁺ ion irradiation

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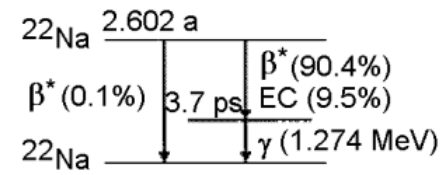
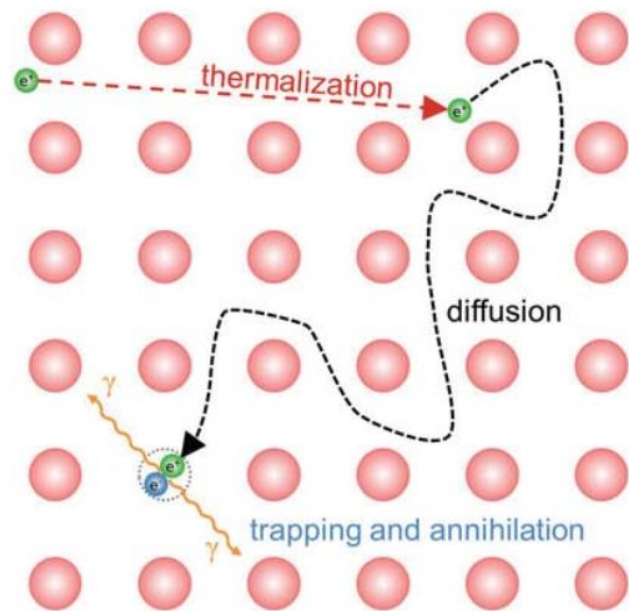
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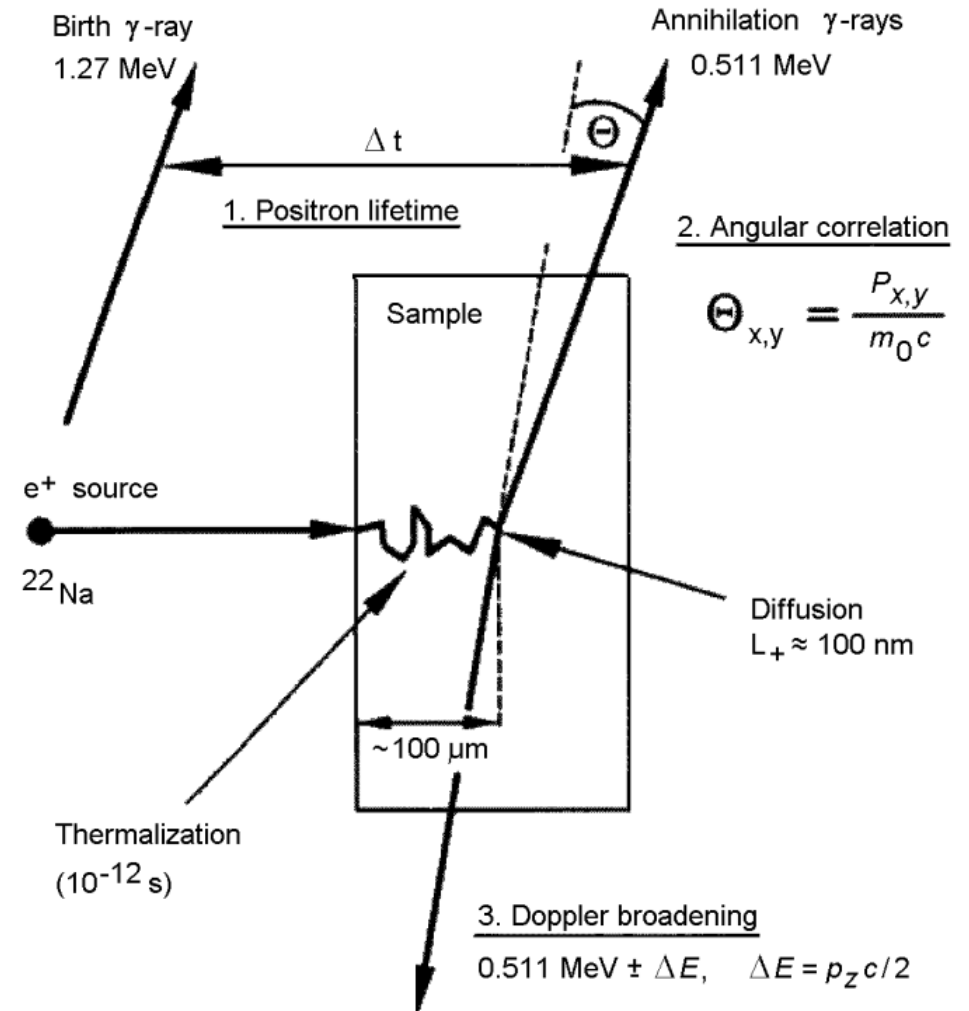


Principio de detección de defectos en PAS

Esquema de los procesos posteriores a la implantación del e^+

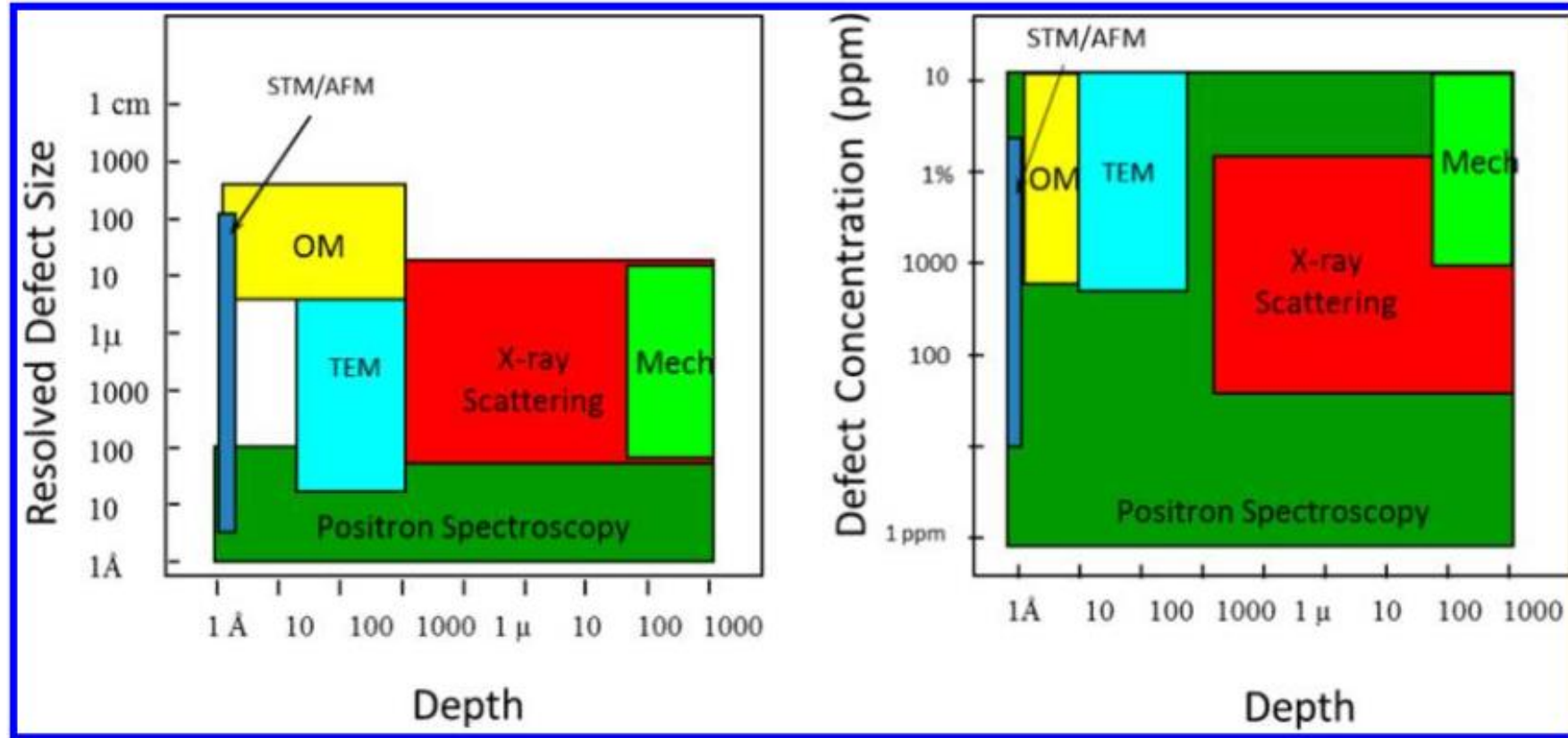


Bases de PAS



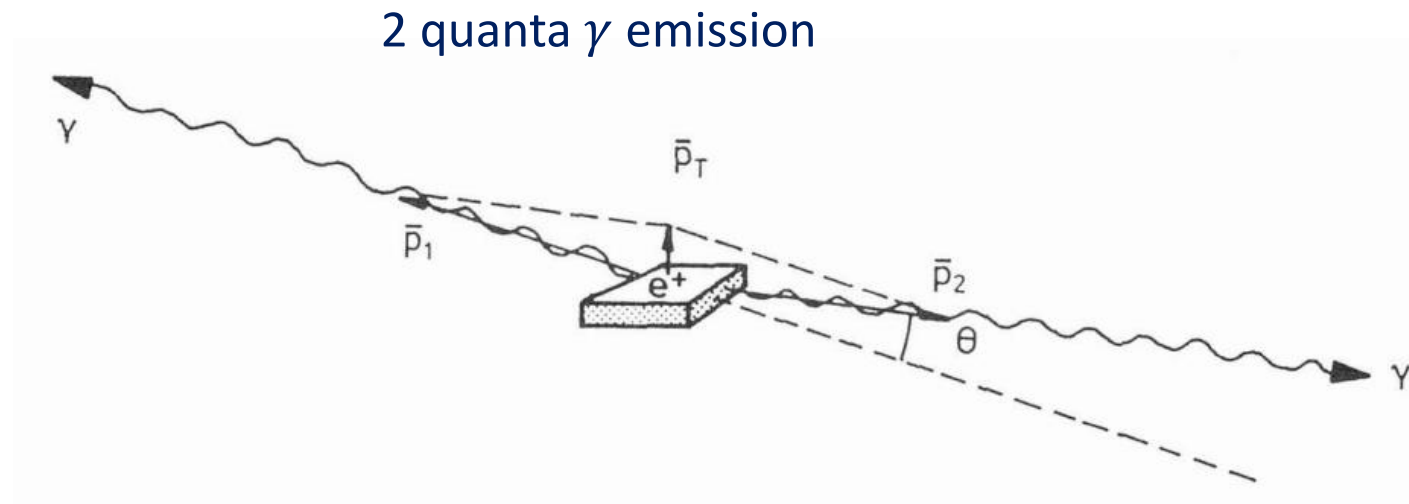
PAS advantages

- It is the only technique that can identify vacancies of atomic dimensions and provides information about:
 - size,
 - structure and
 - concentration (10^{-7}).
- not destructive,
- not sensitive to lattice parameter variations,
- does not measure the effect that defects produce on the properties of materials and
- higher resolution than TEM.



TEM: transmission electron microscopy, **STM:** scanning tunneling microscopy, **AFM:** atomic force microscopy, **OM:** optical microscopy, **Mech:** mechanical method

5. Introduction to PAS. Positron. History



1928: Dirac predicts the positron.

1932: Anderson discovers the positron in a cloud chamber.

1940: PAS is used to study:

- Fermi surface in metals and semiconductors.
- Electron density functions.

1960: positron trapping is discovered.

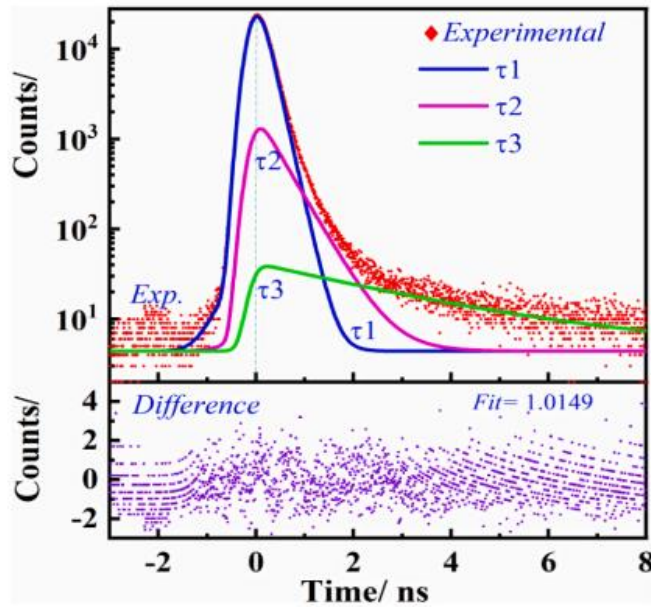
Applications:

- Astrophysics
- Particle physics
- Biomedicine
- Materials Science

Types of PAS

Positron Annihilation Lifetime Spectroscopy (PALS)

$$N(t) = \sum_{i=1}^{k+1} \frac{I_i}{\tau_i} \exp\left(\frac{-t}{\tau_i}\right)$$



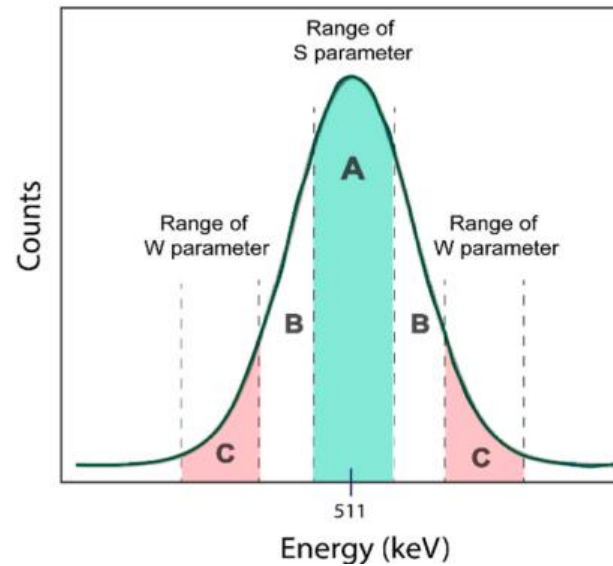
[Selim, 2021]

Softwares: PAFIT [Olsen] y LT [Kansy]

Doppler Broadening Spectroscopy (DBS)

ΔE (Doppler shift)

$$\Delta E = \frac{cp_z}{2}, 511 \text{ eV}$$

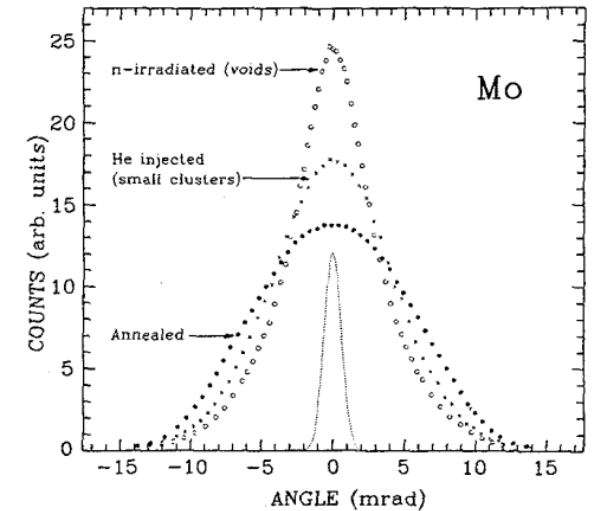


[Selim, 2021]

Software: SP [Dryzek]

Angular Correlation of Annihilation Spectroscopy (ACAR)

$$\Delta\theta_{x,y} = \frac{\Delta\theta}{p_{x,y}}, 511 \text{ eV}$$



[Eldrup, 1995]

Tipos de Espectroscopía de aniquilación positrónica (PAS)

Tipos de PAS

Positron Annihilation Lifetime Spectroscopy (PALS)

$$N(t) = \sum_{i=1}^{k+1} \frac{I_i}{\tau_i} \exp\left(\frac{-t}{\tau_i}\right)$$

- Each type of defect has a characteristic positron lifetime.
- Information about the size and structure of the defect.

Doppler broadening spectroscopy (DBS)

$$\Delta E \text{ (Doppler shift)}$$
$$\Delta E = \frac{cp_z}{2}, 511 \text{ keV}$$

- Doesn't provide information about the size of the defects.
- Qualitative information on changes about defect types based on the S-W correlation.

Angular correlation of annihilation radiation (ACAR)

$$\Delta\Theta_{x,y} = \frac{\Delta\Theta}{mc} \frac{p_{x,y}}{511 \text{ keV}}$$

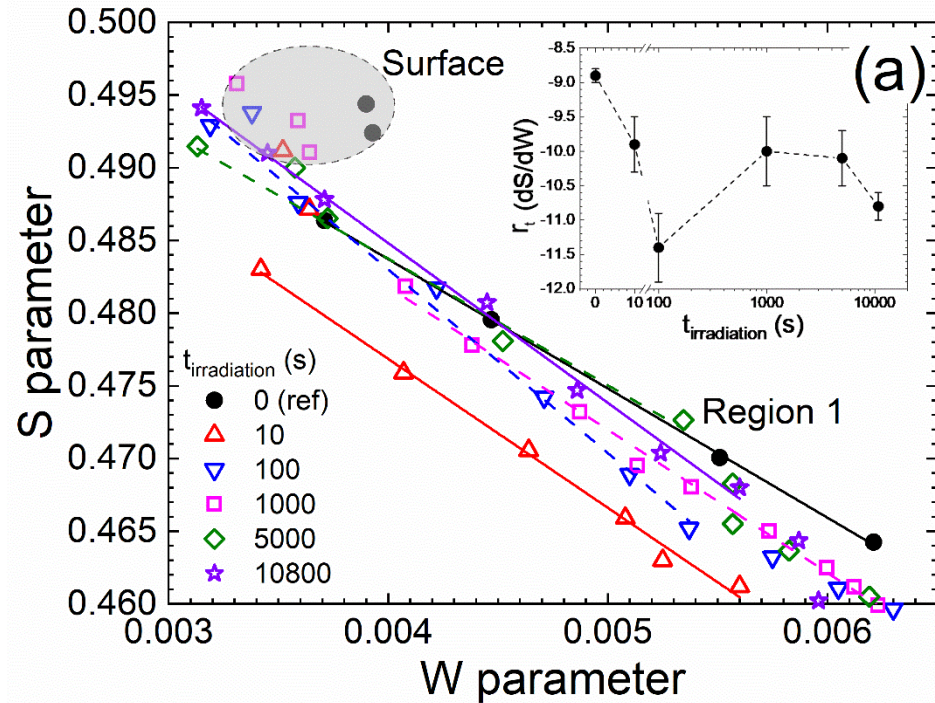
- Information about the electronic structure.
- More detailed information than PALS.
- More complex processing.

Materials and methods. Equipment

Characterization technique	Equipment	Institution	Country
SEM	Tescan Vega	IMRE	Cuba
Raman	Horiba Jobin-Yvon LabRAM, Horiba (iHR320)	IJL,INRS	France, Canada
XRD	Diffractionmeter PANanalytical EMPYREAN Co K _α 1=1.78892 nm	JINR	Russia
FTIR	Bruker Tensor 27	IMRE	Cuba
RBS, Irradiación	Van der Graaff accelerator	JINR	Russia
Positron annihilation (PAS)		JINR	Russia

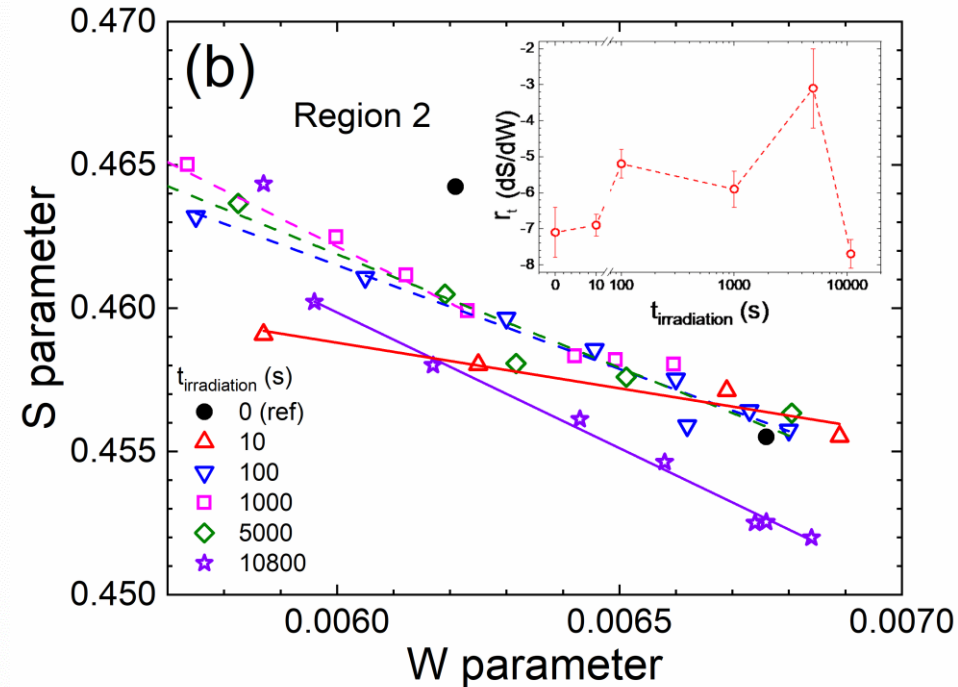
CuO Irradiated nanostructures: DBS

Region 1



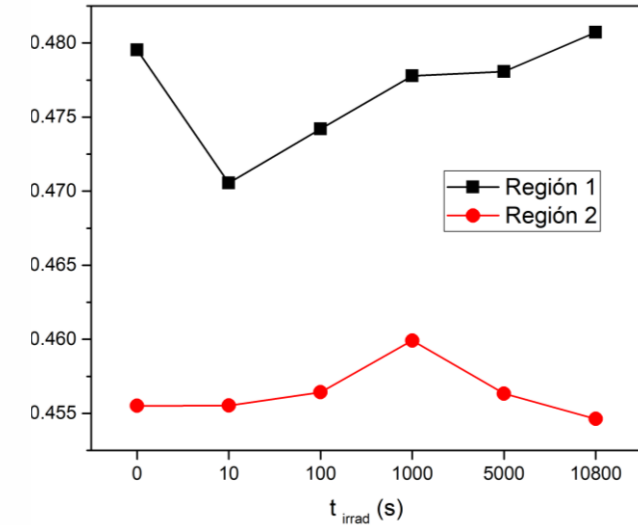
- Inicialmente se forman muchas monovacancias, las cuales se aglomeran y luego forman clústeres.

Region 2



- Los clústeres de monovacancias se disocian formando defectos complejos $\text{Cu}-V_O^+$ o $\text{Cu}-V_{Cu}^-$.

Parámetro S y t_{irrad}



- En la **Región 1** la concentración de las monovacancias aumenta continuamente.
- En la **Región 2**, S solo disminuye luego de 1000s sugiriendo la reducción por los que las dimensiones de los clústeres de defectos complejos y la recuperación de la red.

6. Referencias bibliográficas
