

# Квантово-химические расчеты димеров оганесона и его гомологов

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# Motivation

Determining the limits of applicability of the Mendeleev's periodic law:

- Investigation of properties in comparison with homologues in the noble gases group.
- Investigation of the relativistic effects contribution to molecular properties.
- 2-particle interaction potentials for molecular dynamics methods → thermodynamic properties.

# Methods

## Hamiltonian:

Relativistic → Non-Relativistic with pseudopotential

Mosyagin N.S., Zaitsevskii A.V., Titov A.V. (2020)

GRECP - Generalized Relativistic Effective Core Potential:

For Og atom:

- replaces 92 electrons with pseudopotential leaving us overall with a problem for the  $6s^26p^66d^{10}7s^27p^6$  configuration,
- allows to take into account relativistic effects and Breit interaction.

## Electronic correlation:

Coupled-cluster method – CCSD(T)

Programs - DIRAC, CFOUR

# Basis set

Gaussian Type Orbitals:

$$R_l^\alpha = N_l^\alpha r^l e^{-\alpha r^2}, \quad l = 0(s), 1(p), 2(d) \dots$$

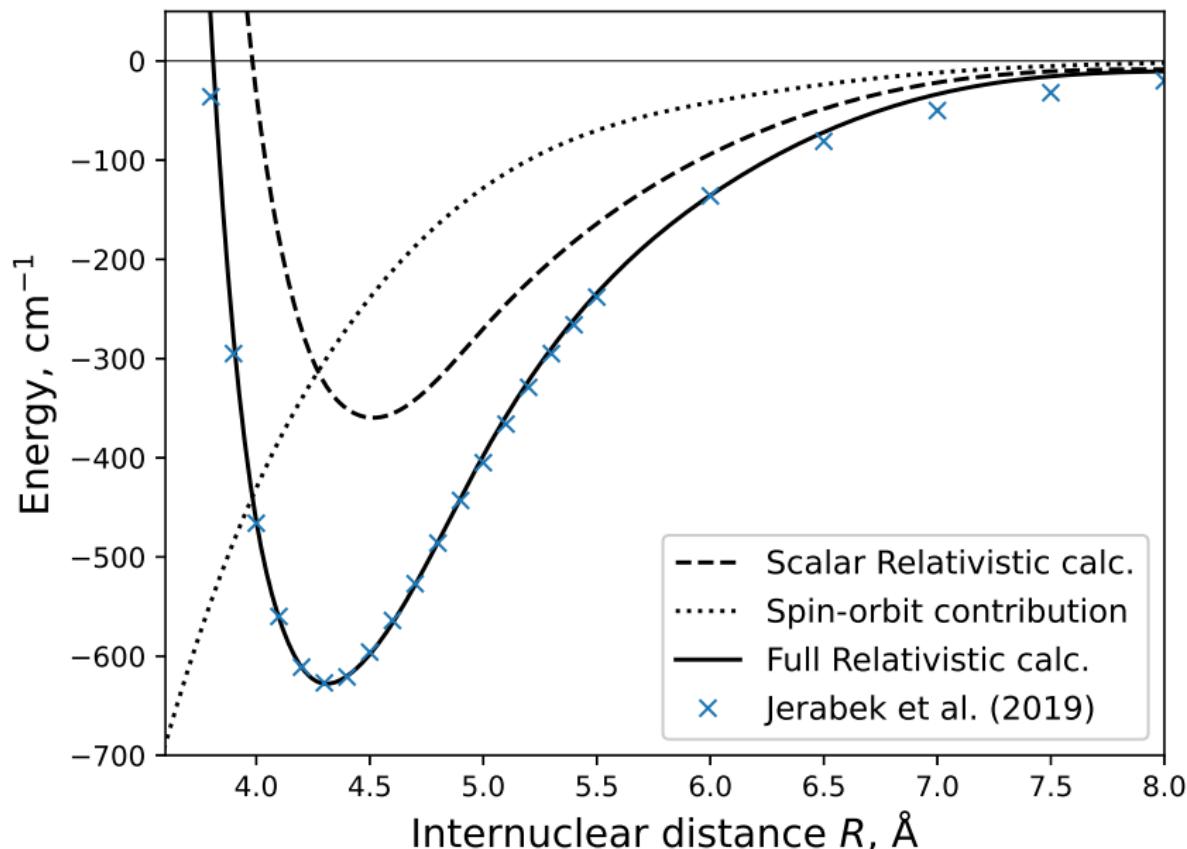
Kaygorodov M. Y. et al. //Physical Review A. – (2021):  
19s17p14d12f4g3h3i – basis set optimized on FSCC-SD  
calculations of electron affinity of Og.

Other approaches:

Jerabek P. et al. (2019): Construction of several smaller basis sets  
and extrapolation to complete basis set.

Shee A., Knecht S., Saue T. (2015); de Macedo L. G. M. et al.  
(2023) – Standard basis sets.

# Og<sub>2</sub> potential energy curve



# Og<sub>2</sub> molecular properties

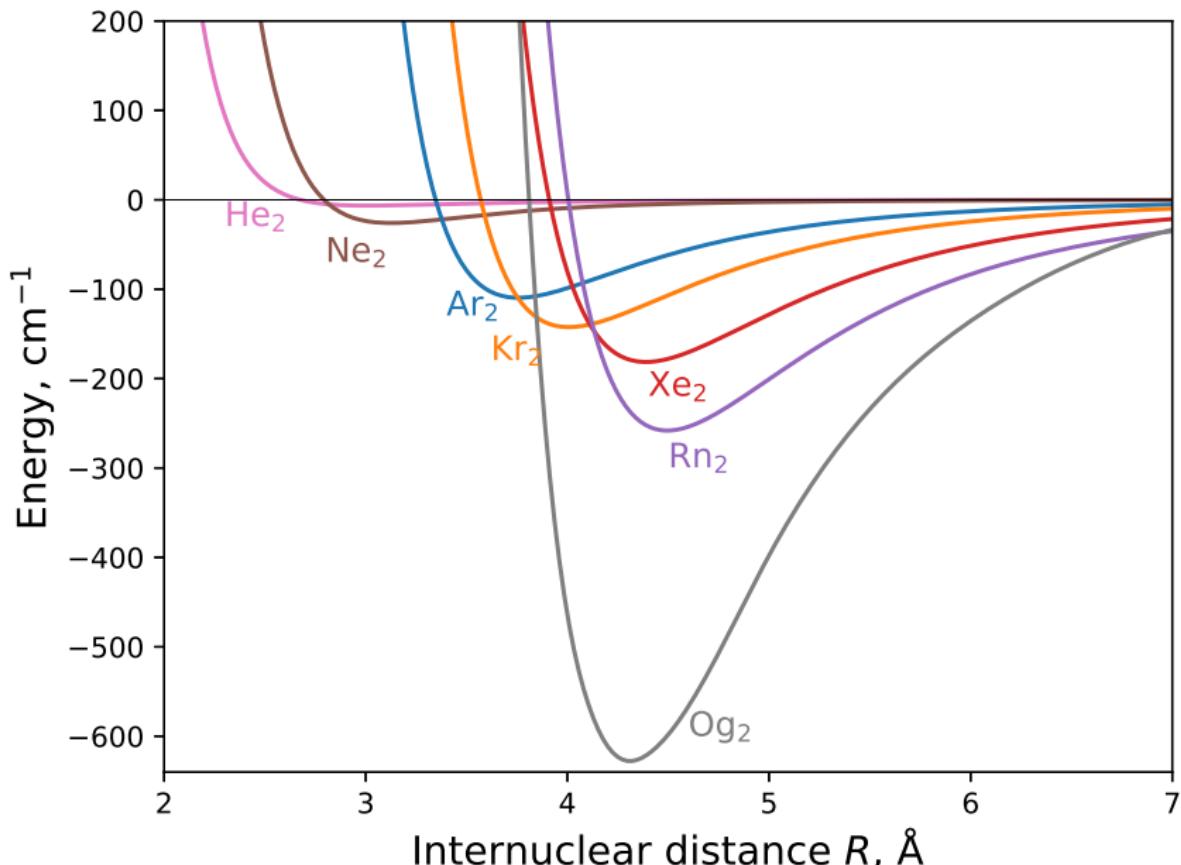
|   | $D_e$ , cm <sup>-1</sup> | $R_e$ , Å |
|---|--------------------------|-----------|
| CCSD(T) Scalar-Relativistic               | 360                      | 4.515     |
| CCSD(T) Full Relativistic                 | 628                      | 4.311     |
| CCSDT(Q) Scalar-Relativistic <sup>1</sup> | 366.17                   | 4.4923    |
| CCSDT(Q) Full Relativistic <sup>1</sup>   | 626.69                   | 4.3137    |
| CCSD(T) <sup>2</sup>                      | 624.24                   | 4.329     |
| MP2-srLDA <sup>3</sup>                    | 843.50                   | 4.25      |

<sup>1</sup>Jerabek P. et al. (2019)

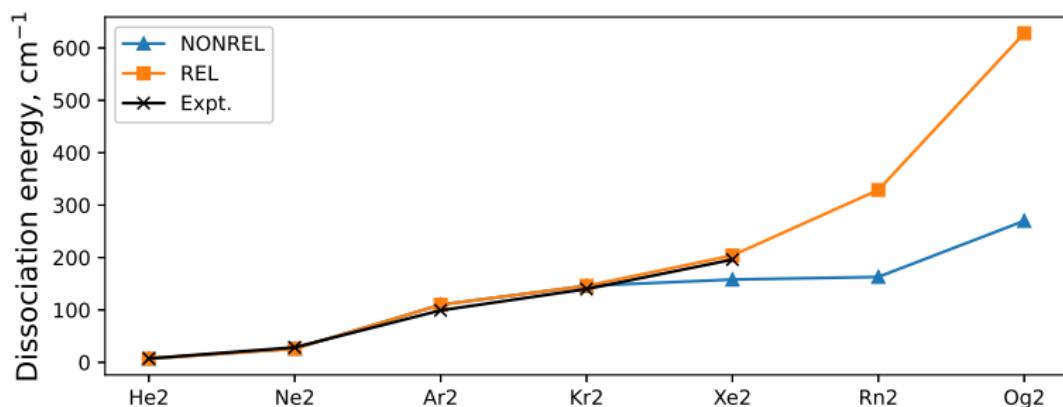
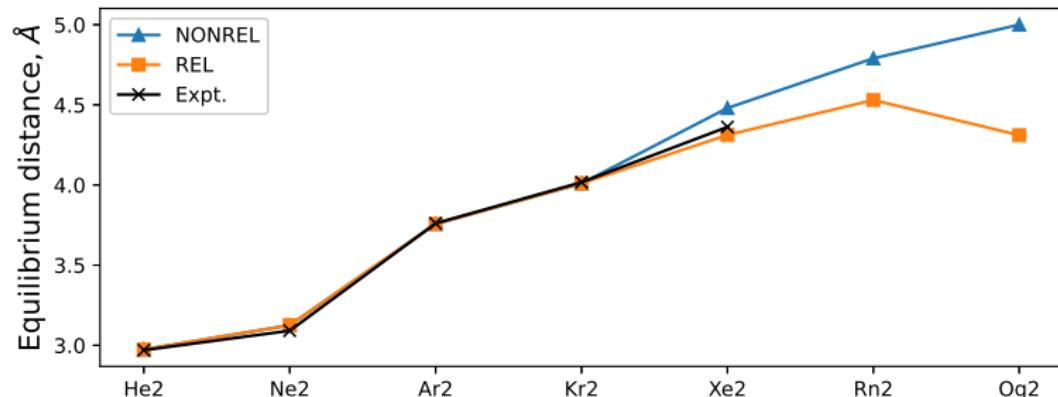
<sup>2</sup>Shee A., Knecht S., Saue T. (2015)

<sup>3</sup>de Macedo L. G. M. et al. (2023)

# Noble gas dimers: potential energies

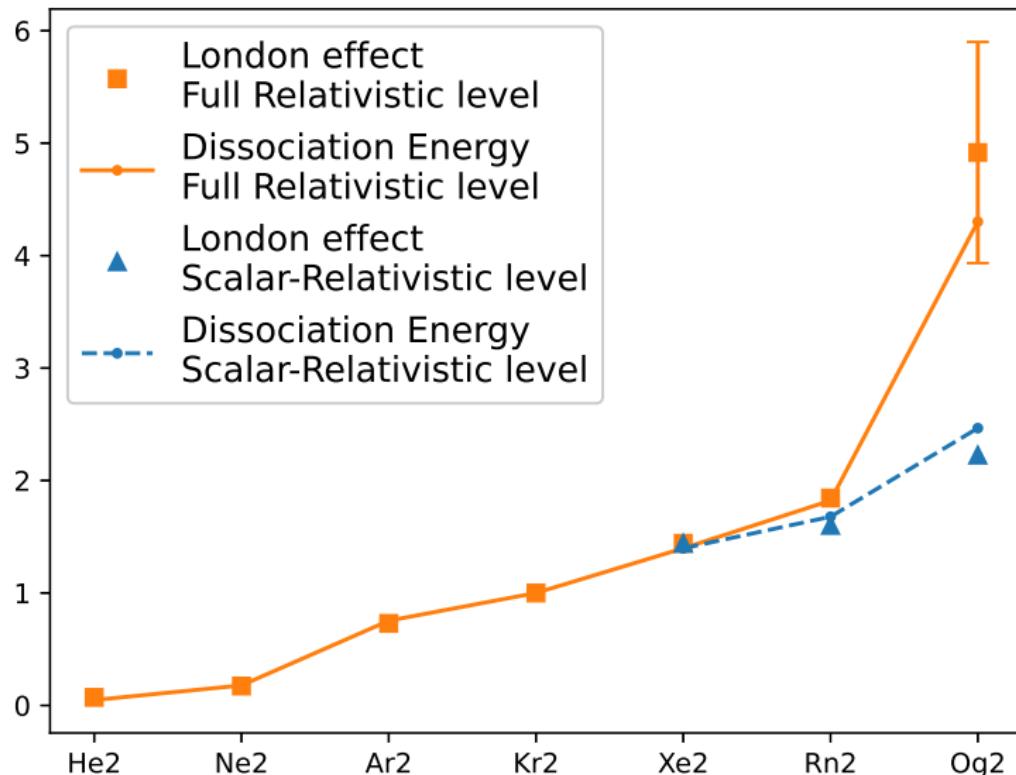


# Noble gas dimers: molecular properties



# London effect $\frac{I\alpha^2}{R^6}$

$I$  – ionization potential,  $\alpha$  – polarizability.



Thank you for your attention